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S Supporting Information

[AB](#page-16-0)STRACT: [Benzo-fused](#page-16-0) nitrogen-containing heterocycles are abundant in biologically active compounds. One of the most important methods for preparing such heterocycles is the $(3 + 2)$ cycloaddition reaction of benzynes with 1,3-dipolar compounds. However, the reactions of unsymmetrically substituted benzynes generally show low selectivity and hence yield mixtures of two regioisomers. In this paper, we describe the synthesis of both regioisomers of multisubstituted

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benzo-fused azole derivatives such as benzotriazoles, 1H-indazoles, and benzo $[d]$ isoxazoles through the regiocomplementary $(3 +$ 2) cycloaddition reactions of 3-boryl- and 3-silylbenzynes with 1,3-dipoles. The improved generation of 3-borylbenzynes from new precursors was one of the most important results of this work, which produced the successful (3 + 2) cycloaddition reactions with exclusive and proximal selectivities. On the other hand, similar reactions of 3-silylbenzynes selectively afforded distal cycloadducts. Analysis of the reaction pathways of these amazing regioselectivities by density functional theory calculations revealed that the $(3 + 2)$ cycloadditions of borylbenzynes are controlled by the electrostatic effect of the boryl group, while those of silylbenzynes are controlled mainly by the steric effect of the bulky silyl groups that produced electrostatically unfavorable adducts via anomalous transition states.

ENTRODUCTION

Benzo-fused azole derivatives are one of the most important classes of nitrogen-containing heterocycles.¹ For example, they have been reported to have a wide range of biological activities such as anticancer,² antidepressant,³ anti[di](#page-16-0)abetic,⁴ antimicrobial, 5 antifungal, 6 antitubercular, 7 and antilipolytic activities. 8 These compounds [h](#page-16-0)ave traditionall[y](#page-16-0) been synthe[siz](#page-16-0)ed using a line[ar](#page-16-0), stepwise [tr](#page-16-0)ansformation [of](#page-16-0) substituted benzene deriv[a](#page-16-0)tives.⁹ However, these methods usually involve many steps and are less effective for the preparation of drug candidate libraries beca[us](#page-16-0)e each compound must be synthesized from a different starting material.

Cycloadditions of benzynes with 4π -components have attracted much attention recently. Among them, the $(3 + 2)$ cycloaddition of benzynes with 1,3-dipole compounds has served as an effective alternative route to benzo-fused azole derivatives.¹⁰ In 2007, Yamamoto et al. reported a synthesis of 1H-indazole^{10b} using the $(3 + 2)$ cycloaddition of benzynes, generated [fro](#page-17-0)m 2-(trimethylsilyl)phenyl triflate with a fluoride source, to d[iazo](#page-17-0) compounds. Subsequently, a large number of benzo-fused azole derivatives such as 1H-indazole,^{10d,h,j} 2Hindazole, $^{10\mathrm{k}}$ benzotriazole, $^{10\mathrm{c,f,g}}$ and benzo $[d]$ isoxazole $^{10\mathrm{a,e,i,l}}$ were synthesized using benzyne reactions. Th[ese t](#page-17-0)ransformatio[ns](#page-17-0) are valuable [for](#page-17-0) convergent diversity-ori[ented](#page-17-0)

syntheses of heterocyclic compounds and allow effective preparation of drug candidate libraries.

However, like most benzyne reactions, ¹¹ the $(3 + 2)$ cycloaddition of unsymmetrically substituted benzynes with 1,3-dipoles is not always regioselective. [Fo](#page-17-0)r example, the reaction of 3-methylbenzyne 1A with 4-methoxyphenyl azide 6a yields a 1:1 mixture of two regioisomers of benzotriazole (proximal- and distal-7Aa) (Table 1, entry 1), 10g and a similar reaction of 3-fluorobenzyne 2A also exhibited poor regioselectivity (entry 2).^{10g} The reaction [o](#page-1-0)f 3-meth[oxyb](#page-17-0)enzyne $3A$ is one of the special cases in which the $(3 + 2)$ cycloaddition with 6a exclusively g[ave](#page-17-0) *distal*-9Aa (entry 3), 10g and similar regioselective $(3 + 2)$ reactions were also reported using 3alkoxybenzynes.^{10b,c,e,g,h,j} In addition, Suzuki [et](#page-17-0) al. discovered another kind of the regioselective $(3 + 2)$ cycloaddition reaction using [\(3-trialky](#page-17-0)lsilyl)benzyne with nitrone.^{10a} However, since then only a few groups have focused on the regiocontrol of the $(3 + 2)$ cycloadditions of [be](#page-17-0)nzynes possessing a substituent other than an alkoxy group,^{10e,12,13} although so many new 1,3-dipolar cycloadditions have been reported over the last dacade.¹⁰ Moreover, 3-silylbenzyn[es](#page-17-0) [have](#page-17-0)

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Table 1. Regioselectivities of $(3 + 2)$ Cycloadditions of 3-Substituted Benzynes 1A−5A with 4-Methoxyphenyl Azide 6a

been generated under harsh reaction conditions using n-BuLi.^{10a,e,12}

We recently reported that 3-boryl- and 3-silylbenzynes unde[rgo D](#page-17-0)iels−Alder reactions with substituted furans and pyrroles to provide cycloadducts with high distal regioselectivity.¹⁴ We applied these benzynes to the $(3 + 2)$ cycloadditions with various 1,3-dipoles and discovered the following. The cy[clo](#page-17-0)addition reactions of 3-borylbenzynes 4 selectively gave proximal benzo-fused azoles 10, whereas similar reactions of 3 silylbenzynes 5 gave distal products 11. These opposite regioselectivities were also generally obtained with a range of 1,3-dipoles (typical examples in Table 1, entries 4 and 5; for more examples, see Table 3). We applied these protocols to the regioselective synthesis of the biologically interesting triazole analogue of hippadine, vo[ro](#page-3-0)zole, and its regioisomer. To reveal the origin of the opposite regioselectivities of the $(3 + 2)$

cycloadditions of 4 and those of 5, we performed theoretical analysis of the reaction pathways by density functional theory (DFT) calculations. Although a part of this work was reported in our recent brief account, 15 this paper describes the full details of this work. In particular, it provides new information about the reasons for the improv[em](#page-17-0)ent of the borylbenzyne precursor and its development process, the experimental procedures and spectroscopic data of the benzyne precursors and products, the preparation of an aza analog of hippadine, and a full discussion about regioselectivities. Our previously published account only summarized the results without discussing them in detail, and in this paper we discuss the details of each reaction, pointing out many important features of the reactions that were not mentioned in the previous account.

■ RESULTS

Development of Improved Preparation Methods of 3- Boryl- and 3-Silylbenzynes. The most critical problems addressed in this project are the development of better precursors for the boryl- and silylbenzynes (4 and 5) and the optimization of the conditions for benzyne generation followed by $(3 + 2)$ cycloaddition, because our preliminary trials, in which the generation methods for 4 and 5 (developed in our previous Diels–Alder reactions^{14b}) were applied to the $(3 + 2)$ cycloaddition with 4-(tert-butyl)phenyl azide 6b, were not very successful. For example, i-PrM[gC](#page-17-0)l·LiCl (1.2 equiv) was added to an $Et₂O$ solution of 6-boryl-2-iodo-4-methylphenyl triflate 12B and 6b at −78 °C, and the reaction mixture was stirred at −78 °C for 30 min to generate 4B and promote the cycloaddition. We observed the formation of a cycloaddition product, proximal-10Bb as a single regioisomer; however, its yield was as low as 15%, and a significant amount of byproduct 15 (29%) along with the recovery of 12B (26%) were observed (Table 2, entry 1). We suspected that 15 was produced by the competitive hydride reduction¹⁶ of 4B by another *i*-PrMgCl molecule, which would leave behind a substantial amount of 12B.

Therefore, to improve the yield of proximal-10Bb, we attempted to suppress the formation of 15 using two approaches. First, we used Grignard reagents without $β$ -

Table 2. Efficient Preparation of 3-Borylbenzyne 4B for (3 + 2) Cycloaddition to 4-(tert-Butyl)phenyl Azide 6b

 a The yield of *proximal*-10Bb was determined by ¹H NMR; *distal*-10Bb was not detected in a crude product of each entry. b The formation of 15 (29% yield) and the recovery of 12B (26% yield) were also observed. "Isolated yield of *proximal*-10Bb.

Figure 1. Structure of 1,3-dipoles 6a−6s.

hydrogen atoms, such as MeMgCl, BnMgCl, and t-Bu₂CHMgCl, instead of *i*-PrMgCl; however, there was no improvement (see Table S1 in Supporting Information). We next examined leaving groups other than −OSO₂CF₃. Knochel et al. reported a very impo[rtant method for prep](#page-16-0)aring functionalized benzynes from 2-iodophenyl 4-chlorobenzenesulfonate using the reaction with i -PrMgCl.¹⁷ The use of corresponding 4-chlorobenzenesulfonate 13B significantly improved the yield of proximal-10Bb by up t[o 4](#page-17-0)8% (Table 2, entry 2). After extensive studies on the benzyne precursors (see Tables S2 and S3 in Supporting Information), we finally fou[nd](#page-1-0) that 2,4,5-trichlorobenzenesulfonate $14B^{18,19}$ was the best precursor of 4B and that its $(3 + 2)$ cycloaddition with 6b afforded proximal-10Bb in 67% yield (Tab[le 2,](#page-17-0) entry 3). It was proved that the iodine−magnesium exchange reaction of 14B proceeded rapidly at -78 °C; however, the subsequent formation of 4B did not proceed at the same temperature when 14B was used as a precursor (see Tables S2 and S3 in Supporting Information). 4B was gradually generated while the reaction mixture was warmed to 0 °C and immediately reacted with 6b. Thus, the leaving ability of the sulfonyloxy group of [the](#page-16-0) [benzyne](#page-16-0) [precursor](#page-16-0) [a](#page-16-0)ffected the rate of benzyne generation, and tuning this leaving ability was the key to suppressing the undesired hydride reduction and allowing the successful reaction of 4 with 6b. The regioisomer distal-10Bb was not detected in any of the cases examined (Table 2), indicating the first successful proximal-selective $(3 + 2)$ cycloaddition of 3substituted benzynes with azides.

Previously, we prepared silylbenzynes 5 thr[ou](#page-1-0)gh the reaction of 2-bromo-6-silylphenyl triflates and n-BuLi and applied them in regioselective Diels−Alder reactions.14a In this study, we found the generation of 5 from 2,6-bis(trimethylsilyl)phenyl triflates 13 using Bu₄NF^{14c,20} was more [suit](#page-17-0)able for the $(3 + 2)$ cycloaddition of 5 because of the milder reaction conditions required. In fact, the rea[ction](#page-17-0) of the in situ generated 5-methyl-3-(trimethylsilyl)benzyne 5B with 6b selectively yielded distal-11Bb (distal: proximal = 10:1) (Table 3, entry 1).¹² It is worth noting that the regioselectivity of 5B was opposite to that of the above-mentioned borylbenzyne 4B. [Th](#page-3-0)is cannot [be](#page-17-0) explained by the electrostatic effects of boron and silicon, which are both less electronegative than carbon (Allred-Rochow electronegativities: B, 2.0; Si, 1.7; C, 2.5).²¹

Scope and Limitation of $(3 + 2)$ Cycloadditions of Boryl- and Silylbenzynes with [Var](#page-17-0)ious 1,3-Dipoles.¹⁵ We next investigated the $(3 + 2)$ cycloadditions of 3-borylbenzynes (4A and 4B) and 3-silylbenzynes 5A−5D with diver[se](#page-17-0) 1,3 dipoles including azides 6a−6m, diazo compounds 6n−6q,

nitrone 6r, and nitrile oxide 6s (Figure 1 and Table 3). The (3 + 2) cycloaddition of 4 generally afforded proximal-10, while that of 5 selectively afforded distal-11. Many e[xam](#page-3-0)ples of opposite selectivities are presented in entries 1−9 of Table 3. On the other hand, nitrone $6r^{13}$ underwent cycloaddition with both 4 and 5 to give distal 2,3-dihydrobenzo $[d]$ isoxazol[es](#page-3-0) (10Br and 11Br) with excell[ent](#page-17-0) selectivity (entry 10; similar reactions of 5 with 6r were previously reported by Suzuki^{10a} and Danishefsky^{10e} under harsh conditions). This is the only exception in which both 4 and 5 showed the same di[stal](#page-17-0) selectivity.

The $(3 + 2)$ cycloaddition of 4 with Me₃SiCHN₂ 6n exclusively gave a desilylated indazole, proximal-10Bn (entry 9), while similar reactions of 5 with diazo compounds (6p and 6q) gave distal 1H-indazoles (11Bp and 11 Bq) (entries 18 and 19). Silylbenzyne 5B and benzonitrile oxide 6s were simultaneously generated from their precursors (16B and Nhydroxybenzimidoyl chloride, respectively) using Bu_4NF in a single flask and exclusively gave distal benzo $[d]$ isoxazole 11Bs (entry 20).

The following features are also worth noting: (i) Functional groups such as olefin $(6i)$, ester $(6g$ and $6l)$, bromo $(6d)$, iodo (6k), and cyano (6m) groups were tolerated under the reaction conditions when either i-PrMgCl·LiCl (for generating 4) or a fluoride source (for generating 5) was used. (ii) The reactions between 4 and most of the azides, including the bulky diphenylmethyl azide 6h, resulted in complete proximal selectivity, and the reaction with the extremely bulky adamantyl azide 6j also exhibited slight proximal preference (proximal-10Bj:distal-10Bj = 1.1:1) (Table 3, entry 8). (iii) The use of 5B′, which has the more bulky tert-butyldimethylsilyl group, resulted in better distal selectivity [\(](#page-3-0)distal-11Be′:proximal-11Be′ $= 10:1$) than did the use of 3-(trimethylsilyl)benzyne **5B** (entry 4). Similar results were also observed in the reaction with 6n (entry 9). The distal selectivities also increased with increasing steric bulk of the alkyl chains in the azides (6e, 6h, and 6j; compare entries 4, 7, and 8). (iv) The substituent R^1 (H, Me, F, Cl) at the C5 position of 4 and 5 had little effect on the selectivity of the reaction with **6e** (entries 3, 4, 16, and 17).

Synthetic Applications of the $(3 + 2)$ Cycloadducts. The proximal-selective cycloaddition of 3-borylbenzynes 4 was applied to the synthesis of triazole analogues (18A and 18B) of a testicular function inhibitor, hippadine 19^{22} (Scheme 1a). When a suitably functionalized azide derivative 6t was used, the $(3 + 2)$ cycloaddition of borylbenzynes, ge[ner](#page-17-0)ated from [t](#page-4-0)he corresponding precursors (14A and 14B), gave benzotriazoles (proximal-10At and proximal-10Bt) in 69% and 67% isolated yields, respectively, with exclusive proximal selectivities. Their intramolecular Suzuki−Miyaura coupling afforded 17A and 17B, which were oxidized by $MnO₂$ to give 18A and 18B in 53% and 61% isolated yields, respectively, in two steps.

The formal synthesis of vorozole 21^9 (Scheme 1b) was performed using the distal-selective cycloaddition of 3 silylbenzynes 5. First, 3-(tert-butyld[im](#page-16-0)ethylsilyl)[be](#page-4-0)nzyne, generated from 16E′ ²³ by Bu4NF, reacted with trimethylsilylmethyl azide [6u](#page-17-0) to give distal-11Eu′ (84% isolated yield) along with proximal-[11](#page-17-0)Eu′ (6% yield). Desilylation of distal-11Eu′, followed by deacetalization, afforded 20 (90% overall yield), a key intermediate for the synthesis of biologically active compounds such as aromatase inhibitor 21⁹ and a PI3K inhibitor.²⁴ Moreover, we synthesized a regioisomer 23 of vorozole 21 via the proximal-selective $(3 + 2)$ [cy](#page-16-0)cloaddition of a 3-boryl[be](#page-17-0)nzyne, generated from 14C, with 6u (Scheme 1c).

Table 3. Complementary Regioselective $(3 + 2)$ Cycloaddition of Borylbenzynes 4 and Silylbenzynes 5

16A, 5A, 11A: $R^1 = H$, $\text{SIR}_3 = \text{SIMe}_3$; **16B, 5B, 11B:** $R^1 = \text{Me}$, $\text{SIR}_3 = \text{SIMe}_3$
16B', 5B', 11B'; $R^1 = \text{Me}$, $\text{SIR}_3 = \text{Si}(t\text{-Bu})\text{Me}_2$; **16C, 5C, 11C:** $R^1 = F$, $\text{SIR}_3 = \text{SiMe}_3$ **16D, 5D, 11D:** R^1 = Cl, Si R_3 = SiMe₃

 a Proximal regioisomer 10 was exclusively observed in ${}^1{\rm H}$ NMR spectra of the crude product. ${}^b{\rm Total}$ isolated yield of distal and proximal isomers 11. ${}^c{\rm The}$ ratio of distal to proximal products, determined u The ratio of distal to proximal products, determined using ¹H NMR spectra of the crude product, is shown in brackets. ^{*d*}Results for 3-(tertbutyldimethylsilyl)benzyne 5B' [SiR₃ = Si(t-Bu)Me₂], which was used instead of 3-(trimethylsilyl)benzyne 5B (SiR₃ = SiMe₃) and provided a mixture of the corresponding distal- and proximal-11B' [SiR₃ = Si(t-Bu)Me₂]. The ratio of distal- to proximal-11B' [SiR₃ = Si(t-Bu)Me₂] is shown in brackets. "Only the corresponding desilylated product was isolated for *proximal-11*. This fall regioisomer 10Br was exclusively observed in ¹H NMR spectra of the crude product. $Ad = 1$ -adamantyl.

These regiocomplementary syntheses are valuable because the methylation of norvorozole produced a 1:1:1 mixture of 21 and its two regioisomers.²⁵ Our method involves fewer steps and allows the convergent synthesis of benzo-fused azoles from benzyne precursors [and](#page-17-0) 1,3-dipoles. Thus, it should be useful for the combinatorial synthesis of diverse derivatives of biologically active compounds.

■ DISCUSSION

Herein, we discuss why the $(3 + 2)$ cycloadditions of borylbenzynes 4 gave proximal products and those of silylbenzynes 5 gave distal products using the results of density functional theory (DFT) calculations.²⁷

First, we checked the internal angles of the geometryoptimized 3-boryl- (4A and 4B) and [3-](#page-17-0)silylbenzynes (5A−5D, and $5B'$) to evaluate the contribution of distortion energy,¹² and we also analyzed the charge distributions of these benzynes by performing natural population analysis in the isolated syste[m](#page-17-0) (without solvent effect and coordination of metals) (Table 4).²⁸ Two important features of the benzynes were found: (i) The internal angles and the charge distribution of all these benzyn[es](#page-17-0) are almost exactly the same, and the substituent $R¹$ at the C5position of benzyne has little effect. (ii) C2s are more electrophilic than C1s because of the electrostatic effect^{10a,14,15} of the boryl and silyl groups and also because of the benzyne distortion.^{12,29}

We next analyzed the experimental regioselectivities of the (3 + 2) cycl[oadd](#page-17-0)ition reactions between two benzynes (4B and 5B) and three 1,3-dipoles (6e, 6o, and 6r) (for natural charges of these 1,3-dipoles, see Figure 2) to find whether they were consistent with their calculated charge distributions and benzyne distortions. The experimental regioselectivities of the reactions between borylbenzyne 4B and the 1,3-dipoles (6e, 6o, and 6r) are consistent with the calculated results, while those

Table 4. Internal Angle and Natural Charge of 3- Borylbenzynes 4 and 3-Silylbenzynes 5

Figure 2. Natural charges of 1,3-dipolar compounds (6e, 6o, and 6r).

between silylbenzyne 5B and the 1,3-dipoles (6e and 6o) are not (Table 3, entries 4, 9, and 10). These contrasting results clearly indicate that the regioselectivities are accounted for by neither the [e](#page-3-0)lectrostatic effect nor the benzyne distortion.

Table 5. Theoretical and Experimental Ratios in the $(3 + 2)$ Cycloaddition Reactions of Borylbenzynes 4 and Silylbenzyne 5

 a Energy difference between *distal-*TS and *proximal-*TS. Positive $\Delta\Delta H^\ddagger$ or $\Delta\Delta G^\ddagger$ indicates that *distal-*TS is higher than *proximal-*TS, while negative $\Delta\Delta H^{\ddagger}$ or $\Delta\Delta G^{\ddagger}$ represents the opposite. ^bActivation energy difference is shown as a $\Delta\Delta H^{\ddagger}$ value. ^cActivation energy difference is shown as as a $\Delta \Delta G^{\ddagger}$ value.

Therefore, we performed a theoretical analysis of the reaction pathways, including the transition states, to quantitatively rationalize the origin of the selectivities.

The transition states (TS1−TS4) of the reactions between benzyl azide 6e and two benzynes (3-borylbenzyne 4B and 3 silylbenzyne 5B) were obtained as typical cases (Table 5, entries 1 and 2). TS2, which leads to proximal-10Be, has an energy 2.58 kcal/mol lower than that of TS1, which leads to distal-10Be. This energy difference corresponds to 1:>50 proximal selectivity, which is in good agreement with our experimental result (distal-10Be:proximal-10Be = 1:>50; Table 3, entry 4 and Table 5, entry 1). On the other hand, TS3, which leads to distal-11Be, has an energy 1.1 kcal/mol lower than that [o](#page-3-0)f TS4, which leads to proximal-11Be. This difference corresponds to 6.5:1 distal selectivity, which is in reasonable agreement with our experimental result (distal-11Be: proximal-11Be = 3.3:1; Table 3, entry 4 and Table 5, entry 2).³⁰

We also performed similar calculations for thirteen other reactions between [4](#page-3-0)B (or 5B) and various 1,[3-d](#page-17-0)ipolar compounds 6. All of these results are in good agreement with the experimental data (Table 5, entries 4−16). These facts suggest that the calculated transition states are reliable and that their structures can provide valuable information that can be used to identify the origin of the regioselectivities. 30

As a reference, the transition state TS33 of the reaction of benzyne 24A and 6e was also calculated, in which the bond distance (2.40 Å) between the internal nitrogen of 6e and the benzyne carbon is shorter than that (2.77 Å) between the terminal nitrogen and the benzyne carbon (Figure 3a). This is despite the fact that there should be some steric interaction between the benzylic methylene moiety of 6e and [th](#page-6-0)e benzyne hydrogen. The corresponding bond distances of the TS34, derived from 5-methylbenzyne 24B, and 6e are close to those of TS33, which implies that a methyl group at the C5-position of benzyne hardly affects the structure of the transition state. The lengths of the two bonds forming in TS2, which is more stable than TS1 because the electrostatic combination of the reacting atoms is well matched (Figure 3b), are also very similar to those of TS34. These facts imply that there is little steric interaction between the boryl group a[nd](#page-6-0) the benzyl substituent and that the exclusive formation of *proximal*-10Be is dominated by the electrostatic interactions between the two reactants and also by the benzyne distortion, both of which are caused by the boryl group.¹⁵

In the case of silylbenzyne 5B, the distance between the internal nitr[og](#page-17-0)en and the benzyne C2 is much longer in TS4 (2.51 Å) than in TS34 (2.41 Å) because of the strong steric interaction between the trimethylsilyl group and the benzyl group. This makes the electrostatically and distortionally

Figure 3. Transition states of $(3 + 2)$ cycloaddition reactions of benzynes with 1,3-dipoles.^{15,32} (a) Transition states of the reactions between benzyne 24A and 6e (TS33) and between 5-methylbenzyne 24B and 6e (TS34). (b) Most probable transition states (TS1 and TS2) of the reaction between 3-boryl-5-methylbenzyne 4B and benzyl azide 6e giving distal-10[Be](#page-17-0) and proximal-10Be[,](#page-17-0) respectively. (c) Most probable transition states (TS3 and TS4) of the reaction between 5-methyl-3-(trimethylsilyl)benzyne 5B and 6e giving distal-11Be and proximal-11Be, respectively. (d) Most probable transition states (TS5 and TS6) of the reaction between 5-methyl-3-silylbenzyne 5B'' and 6e.

favorable TS4 less stable than the unfavorable TS3 (Figure 3c) and results in the preferential formation of distal-11Be. We also analyzed two transition states (TS5 and TS6) derived from a virtual benzyne 5B′′ possessing a SiH3 group (Figure 3d). The distance between the internal nitrogen and the benzyne C2 of **TS6** (2.41 Å) is shorter than that of **TS4** (2.51 Å) but similar to that of TS34 (2.41 Å). Although the benzyne distortion of 5B′′ is quite similar to that of 5B (Table 4), the energy difference between TS5 and TS6 is smaller than that between TS3 and TS4 (Figure 3c and d). These results [a](#page-4-0)lso prove that the steric bulkiness of the silyl groups has a significant adverse influence on the electrostatically favorable proximal transition state. This steric influence even overrides the attractive electrostatic interaction and the benzyne distortion.^{15,31}

■ **CONCLUSIONS**

In conclusion, we have achieved complementary regiocontrol of the $(3 + 2)$ cycloaddition reaction of benzynes with 1,3-dipoles by the unique substituent effects of the boryl and silyl groups. Both regioisomers of benzo-fused azoles such as benzotriazole, 1H-indazole, and benzo $[d]$ isoxazole can be prepared by choosing a boryl or silyl group as the benzyne substituent. In particular, the finding of an improved generation method of 3 borylbenzynes 4 from new precursors 14 was one of the most important new results of this work and led to the successful (3 + 2) cycloaddition reactions to demonstrate the first proximalselective examples of the cycloaddition reaction of 3-substituted benzynes with 1,3-dipoles. We have clearly and quantitatively explained the regioselectivities of these reactions by analyzing the transition state structures obtained by DFT calculations. Namely, the $(3 + 2)$ cycloaddition of borylbenzynes 4 is more

electrostatically controlled, while that of silylbenzynes 5 is more sterically dominated. Because the boryl and silyl groups of the cycloaddition products can be converted into carbon, nitrogen, and oxygen substituents as well as a hydrogen atom, 33 the use of 3-boryl- and 3-silylbenzynes would be the ideal solution to the longstanding problem of the regioselectivity of s[ub](#page-17-0)stituted benzynes. Application of this chemistry to the preparation of a wide range of biologically interesting compounds and further improvement of the selectivity are in progress in our laboratory.

EXPERIMENTAL SECTION

General. All reactions were carried out under an argon or nitrogen atmosphere in an oven-dried flask containing a stir-bar with a rubber septum or an inlet adapter with a three-way stopcock. 1-Azido-4 methoxybenzene 6a, ³⁴ 1-azido-4-(tert-butyl)benzene 6b, ³⁵ 2-azido-1,4 dimethoxybenzene 6c, 10c 2-azido-1,3-dibromobenzene 6d, ³⁶ 1-azidodecane $6f$,³⁷ ethyl [2-a](#page-17-0)zidoacetate $6g$,^{10c} (azidomethyl[en](#page-17-0)e)dibenzene $6h, ^{38}$ (E)-(3-azidopro[p-1-](#page-17-0)en-1-yl)benzene $6i, ^{10c}$ (1R,3R,5S[\)-](#page-17-0)1-azidoadamantane $6j$, $10c$ 1-(azidomethyl)-2-[iod](#page-17-0)obenzene $6k$, $10c$ methyl 2(a[zid](#page-17-0)omet[hy](#page-17-0)l) benzoate $6I$,³⁹ 2-(azidomet[hyl](#page-17-0)) benzonitrile $6m$,⁴⁰ \int_0^{π} diazomethyle[ne\)](#page-17-0) dibenzene $6q_r^{41}$ N-hydroxybenzim[ido](#page-17-0)yl chloride (for the in situ preparati[on](#page-17-0) of benzonitrile N-oxide $6s$),⁴² [5-](#page-17-0) (azidomethyl)-6-iodobenzo $[d]$ 1,[3-d](#page-17-0)ioxole $6t, ^{43}$ 2-iodo-6-(4,4,5,5-tetramethyl-1,3-dioxolan-2-yl)phenyl trifluoromethanesulfonate 12A,^{14b} 2iodo-4-methyl-6-(4,4,5,5-tetramethyl-1,3-dio[xol](#page-18-0)an-2-yl)phenyl tri[fl](#page-17-0)uoromethanesulfonate 12B, 14b 2,6-bis(trimethylsilyl)phenyl trifluo[rom](#page-17-0)ethanesulfonate 16A,^{14c} 4-methyl-2,6-bis(trimethylsilyl)phenyl trifluoromethanesulfonate 16B, [14c](#page-17-0) 4-fluoro-2,6-bis(trimethylsilyl)phenyl trifluoromethanes[ul](#page-17-0)fonate $16C_1^{14c}$ and 4-chloro-2,6- bis(trimethylsilyl)pheny[l t](#page-17-0)rifluoromethanesulfonate $16D^{14c}$ were synthesized according to the literature[. Fla](#page-17-0)sh chromatography⁴⁴ was performed with silica gel 60 N, spherical neutral (40–50 μ m). ¹H $\rm NMR$ and $\rm ^{13}C$ $\rm NMR$ spec[t](#page-17-0)ra were recorded on an instrument $(\rm ^{1}H,$ $(\rm ^{1}H,$ 500 MHz; ¹³C, 125 MHz) with chemical shifts reported in ppm relative to

the residual deuterated solvent or the internal standard tetramethylsilane. The high-resolution mass spectra were recorded on an ESI or APCI TOF mass spectrometer. Yield refers to isolated yields of compounds greater than 95% purity as determined by ¹H NMR analysis. ¹H NMR and melting points (where applicable) of all known compounds were taken from references. The regioselectivities were determined by 500 MHz ¹H NMR spectra of crude reaction mixtures. Each regiochemistry of representative cycloaddition products (proximal-10Bb, proximal-10Be, proximal-10Bf, proximal-10Bg, proximal-10Bh, proximal-10 Bi, proximal-10Bj, proximal-10Bn, distal-10Br, distal-11Bc, distal-11Be, distal-11Bf, distal-11Bh, proximal-11Bh, distal-11Bj, distal-11Bn, distal-11Bp, and distal-11Bs) was confirmed by NOESY experiment. The regiochemistries of all other cycloaddition products were predicted by the similarity of ¹H NMR data.

General Procedure I: Proximal-Selective (3 + 2) Cycloadditions of Borylbenzynes 4 (Tables 1−3). An oven-dried flask was charged with a borylbenzyne precursor 12^{14b} −14 (1.0 equiv) and a dipole 6 (3.0 equiv), capped with a rubber septum, then evacuated, and backfilled with argon. Anhydrous $Et₂O$ $Et₂O$ [\(](#page-1-0)[0.10](#page-17-0) M) was added, and the mixture was cooled to −78 °C. Then, a 1.0 or 1.3 M solution of i-PrMgCl·LiCl in THF (1.2 equiv) was slowly added to the reaction mixture over 5 min. After stirring at −78 °C for 30 min, the reaction mixture was warmed to 0 °C, stirred for an additional 30 min, and quenched with a saturated aqueous NH4Cl solution. Then the reaction mixture was extracted with EtOAc, and the aqueous layer was extracted twice with EtOAc. The combined organic layers were washed with a saturated aqueous NaCl solution and dried over anhydrous $Na₂SO₄$. The solution was filtered through a glass filter and concentrated under reduced pressure. The crude product was purified by silica gel flash column chromatography to give boryl benzazole 10.

1-(4-Methoxyphenyl)-7-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-1H-benzo[d]triazole (proximal-10Aa) (Table 1, entry 4). Following the general procedure I, a mixture of $14A^{14b}$ (61 mg, 103 μ mol), 1-azido-4-methoxybenzene $6a^{34}$ (46 mg, 0.31 mmol), and i -PrMgCl·LiCl [\(1](#page-1-0).0 M in THF, 0.12 mL, 0.12 mmol) in Et₂O (1.1 mL, 0.10 M) was stirred for 30 min at 0 °[C.](#page-17-0) The crude product was purified by column chromatography (hexane/EtOAc = $3:1$) to provide the titled compound proximal-10Aa (17 mg, 46%) as a colorless solid. Mp 142−145 °C. ¹H NMR (500 MHz, CDCl₃) δ: 1.08 (12 H, s), 3.89 $(3 H, s)$, 7.03 $(2 H, td, J = 3.0, 9.0 Hz)$, 7.40 $(1 H, dd, J = 7.0, 8.0 Hz)$, 7.41 (2 H, td, $J = 3.0$, 9.0 Hz), 7.92 (1 H, dd, $J = 1.0$, 7.0 Hz), 8.20 (1 H, dd, J = 1.0, 8.0 Hz). ¹³C NMR (125 MHz, CDCl₃) δ : 24.5, 55.6, 83.8, 114.1, 122.7, 123.5, 128.1, 131.3, 136.3, 136.8, 145.3, 160.4. IR (CHCl₃): 1518 cm⁻¹. HRMS calcd for C₁₉H₂₂BN₃NaO₃ (M + Na)⁺ m/z: 374.1652, found 374.1661.

1-[4-(tert-Butyl)phenyl]-5-methyl-7-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-1H-benzo[d]triazole (proximal-10Bb) (Table 2, entry 1). An oven-dried pear-shaped flask was charged with $12B^{14b}$ (44 mg, 90 μ mol), capped with an inlet adapter with a three-way stopcock, then evacuated, and backfilled with argon. Anhydr[ou](#page-1-0)[s E](#page-17-0)t₂O (0.90 mL, 0.10 M) and 1-azido-4-(*tert*-butyl)benzene $6b³¹$ (62 mg, 0.35 mmol) were added via a syringe, and the mixture $(62 \text{ mg}, 0.35 \text{ mmol})$ were added via a syringe, and the mixture was cooled to −78 °C. i-PrMgCl·LiCl (1.3 M in THF, 0.080 mL, 104 μ [mo](#page-17-0)l) was slowly added into the reaction mixture over 5 min. After being stirred at −78 °C for 30 min, the reaction mixture was quenched with a saturated aqueous NH₄Cl solution. The mixture was extracted with EtOAc, and the aqueous layer was extracted twice with EtOAc. The combined organic layers were washed with a saturated aqueous NaCl solution and dried over anhydrous $Na₂SO₄$. The solution was filtered through a glass filter and concentrated under reduced pressure. The product yield and ratio were determined by ¹H NMR [*proximal*-10Bb (15%), 15^{45} (29%), $12B^{14b}$ (26%), 1,4-dimethoxybenzene was used as an internal standard].

Table 2, ent[ry](#page-18-0) 2. Followin[g th](#page-17-0)e general procedure I, a mixture of 13B (25 mg, 48 μ mol), 1-azido-4-(tert-butyl)benzene $6b^{35}$ (23 mg, 131 μ mol), and *i*-PrMgCl·LiCl (1.3 M in THF, 60 μ L, 78 μ mol) in Et₂O (0.[50](#page-1-0) mL, 0.10 M) was stirred for 30 min at 0 °C. [The](#page-17-0) product yield and ratio were determined by ¹H NMR [proximal-**10Bb** (48%), 1,4-dimethoxybenzene was used as an internal standard].

Table 2, entry 3 and Table 3, entry 1. Following the general procedure I, a mixture of $14B$ (21 mg, 34 μ mol), 1-azido-4-(tertbutyl)benzene $6b^{35}$ (20 mg, 113 μ mol), and *i*-PrMgCl·LiCl (1.3 M in THF, 40 μ L, 52 μ mol) in Et₂O (0[.4](#page-3-0)0 mL, 0.10 M) was stirred for 30 min at 0 °C. [T](#page-17-0)he crude product was purified by column chromatography (hexane/EtOAc = 10:1) to provide the titled compound proximal-10Bb (8.9 mg, 67%) as a colorless solid, the regiochemistry of which was determined by NOESY spectrum. Mp 160−165 °C. ¹H NMR (500 MHz, CDCl₃) δ: 1.05 (12 H, s), 1.38 (9 H, s), 2.53 (3 H, s), 7.42 (2 H, d, $J = 8.5$ Hz), 7.53 (2 H, d, $J = 8.5$ Hz), 7.72 (1 H, d, $J = 1.0$ Hz), 7.95 (1 H, d, $J = 1.0$ Hz). ¹³C NMR (125 MHz, CDCl3) δ: 21.1, 24.5, 31.3, 31.3, 34.8, 83.8, 121.6, 125.9, 126.0, 133.4, 134.9, 135.8, 137.9, 146.1, 152.2. IR (CHCl₃): 2968 cm⁻¹. HRMS calcd for $C_{23}H_{31}BN_3O_2 (M + H)^+ m/z$: 392.2509, found 392.2507.

1-(2,5-Dimethoxyphenyl)-5-methyl-7-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-1H-benzo[d]triazole (proximal-10Bc) (Table 3, entry 2). Following the general procedure I, a mixture of 14B (31 mg, 52 μ mol), 2-azido-1,4-dimethoxybenzene 6c^{10c} (50 μ L, 166 μ mol), and *i*-PrMgCl·LiCl (1.3 M in THF, 50 μ L, 65 μ mol) in Et₂O ([0.5](#page-3-0)0 mL, 0.10 M) was stirred for 30 min at 0 °C. [Th](#page-17-0)e crude product was purified by column chromatography (hexane/EtOAc $=$ 10:1) to provide the titled compound proximal-10Bc (13 mg, 65%) as a colorless solid. Mp 116−120 °C. ¹H NMR (500 MHz, CDCl₃) δ : 1.07 (12 H, s), 2.52 (3 H, s), 3.58 (3 H, s), 3.79 (3 H, s), 6.93 (1 H, dd, $J = 1.0$, 9.5 Hz), 7.03–7.06 (2 H, m), 7.76 (1 H, d, $J = 1.0$ Hz), 7.95 (1 H, brs). 13C NMR (125 MHz, CDCl3) δ: 20.6, 24.8, 24.8, 37.3, 83.6, 88.9, 92.5, 114.6, 123.1, 123.7, 128.0, 128.3, 131.4, 134.6, 135.4, 1367.0, 138.4, 145.4. IR (CHCl₃): 1514 cm⁻¹. HRMS calcd for $C_{21}H_{27}BN_3O_4 (M + H)^+$ m/z: 396.2095, found 396.2086.

1-Benzyl-7-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)- 1H-benzo[d]triazole (proximal-10Ae) (Table 3, entry 3). Following the general procedure I, a mixture of 14A (64 mg, 0.11 mmol), (azidomethyl)benzene 6e (41 μ L, 0.32 mmol), and *i*-PrMgCl·LiCl (1.3 M in THF, 0.10 mL, 0.13 mmol[\)](#page-3-0) in Et₂O (1.1) mL) was stirred for 30 min at 0 °C. The crude product was purified by column chromatography (hexane/EtOAc = 10:1) to provide the titled compound proximal-10Ae (32 mg, 89%) as a colorless solid. Mp 89− 92 °C. ¹H NMR (500 MHz, CDCl₃) δ: 1.25 (12 H, s), 6.40 (2 H, s), 7.01 (2 H, d, J = 7.0 Hz), 7.20−7.25 (3 H, m), 7.38 (1 H, dd, J = 7.5 Hz, 8.0 Hz), 8.04 (1 H, dd, J = 1.0 Hz, 7.5 Hz), 8.21 (1 H, dd, J = 1.0 Hz, 8.0 Hz). ¹³C NMR (125 MHz, CDCl₃) δ : 24.6, 52.3, 84.4, 123.3, 123.5, 126.2, 127.3, 128.4, 135.8, 137.1, 137.3, 145.9. IR $(CHCl₃)$: 3010 cm⁻¹. HRMS calcd for C₁₉H₂₃BN₃O₂ (M + H)⁺ m/z: 336.1883, found 336.1863.

1-Benzyl-5-methyl-7-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-1H-benzo[d]triazole (proximal-10Be) (Table 3, entry 4). Following the general procedure I, a mixture of 14B (61 mg, 101 μ mol), (azidomethyl)benzene 6e (40 μ L, 0.30 mmol), and *i*-PrMgCl·LiCl [\(1](#page-3-0).3 M in THF, 90 μ L, 117 μ mol) in Et₂O (1.0 mL, 0.10 M) was stirred for 30 min at 0 $^{\circ}$ C. The crude product was purified by column chromatography (hexane/ $EtOAc = 4:1$) to provide the titled compound proximal-10Be (30 mg, 85%) as a colorless solid, the regiochemistry of which was determined by NOESY spectrum. Mp 116−118 °C. ¹H NMR (500 MHz, CDCl₃) δ: 1.24 (12 H, s), 2.51 (3 H, s), 6.36 (2 H, s), 6.99 (2 H, d, J = 7.0 Hz), 7.17−7.24 (3 H, m), 7.86 (1 H, d, $J = 1.0$ Hz), 7.96 (1 H, brs). ¹³C NMR (125 MHz, CDCl3) δ: 21.1, 24.7, 52.3, 84.4, 122.5, 126.2, 127.2, 128.4, 133.2, 134.4, 137.2, 139.3, 146.7. IR (CHCl₃): 2982 cm⁻¹. HRMS calcd for $C_{20}H_{25}BN_3O_2$ $(M + H)^+$ m/z : 350.2040, found 350.2037.

1-Decyl-5-methyl-7-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-1H-benzo[d]triazole (proximal-10Bf) (Table 3, entry 5). Following the general procedure I, a mixture of 14B (56 mg, 93 μ mol), 1-azidodecane 6f³⁵ (60 mg, 0.33 mmol), and *i*-PrMgCl-LiCl (1.3 M in THF, 85 μ L, 111 μ mol) in Et₂O ([0](#page-3-0).90 mL, 0.10 M) was stirred for 30 min at 0 °[C.](#page-17-0) The crude product was purified by column chromatography (hexane/EtOAc = $4:1$) to provide the titled compound proximal-10Bf (32 mg, 87%) as a colorless solid, the regiochemistry of which was determined by NOESY spectrum. Mp 70−72 °C. ¹ H NMR (500 MHz, CDCl3) δ: 0.86 (3 H, t, J = 7.5 Hz),

1.23−1.37 (14 H, m), 1.40 (12 H, s), 1.87 (2 H, q, J = 7.0 Hz), 2.50 (3 H, s), 5.00 (2 H, t, $J = 7.0$ Hz), 7.85 (1 H, d, $J = 1.5$ Hz), 7.90 (1 H, brs). 13C NMR (125 MHz, CDCl3) δ: 14.1, 21.1, 22.6, 24.9, 26.6, 29.2, 29.3, 29.48, 29.50, 30.9, 31.8, 49.6, 84.3, 122.4, 132.9, 133.9, 139.0, 146.6. IR (CHCl₃): 2927 cm⁻¹. HRMS calcd for C₂₃H₃₉BN₃O₂ (M + H)⁺ m/z : 400.3135, found 400.3156.

Ethyl 2-(5-Methyl-7-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-1H-benzo[d]triazol-1-yl)acetate (proximal-10Bg) (Table 3, entry 6). Following the general procedure I, a mixture of 14B (57 mg, 95 μ mol), ethyl 2-azidoacetate $6g^{10c}$ (35 μ L, 0.30 mmol), and *i*-PrMgCl·LiCl (1.3 M in THF, 90 μ L, 117 μ mol) in Et₂O (1.0 mL, 0.1[0](#page-3-0) M) was stirred for 30 min at 0 °C. [Th](#page-17-0)e crude product was purified by column chromatography (hexane/EtOAc = $3:1$) to provide the titled compound proximal-10Bg (24 mg, 74%) as a colorless solid, the regiochemistry of which was determined by NOESY spectrum. Mp 141−144 °C. ¹H NMR (500 MHz, CDCl₃) δ: 1.21 (3 H, t, J = 7.5 Hz), 1.35 (12 H, s), 2.50 (3 H, s), 4.17 (2 H, q, J = 7.5 Hz), 5.83 (2 H, s), 7.88 (1 H, d, $J = 1.5$ Hz), 7.93 (1 H, brs). ¹³C NMR (125 MHz, CDCl3) δ: 14.0, 21.1, 24.7, 51.3, 61.5, 84.4, 122.6, 133.2, 134.8, 139.4, 146.6, 167.7. IR (CHCl₃): 1755 cm⁻¹. HRMS calcd for C₁₇H₂₅BN₃O₄ $(M + H)^+$ m/z: 346.1938, found 346.1933.

1-Benzhydryl-5-methyl-7-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-1H-benzo[d]triazole (proximal-10Bh) (Table 3, entry 7). Following the general procedure I, a mixture of 14B (72 mg, 119 μ mol), (azidomethylene) d ibenzene $6h^{39}$ (75 mg, 0.36 mmol), and *i*-PrMgCl·LiCl (1.3 M in THF, 0.11 mL, 143 μ mol) in Et₂O ([1.2](#page-3-0) mL, 0.10 M) was stirred for 30 min at 0 °C. [Th](#page-17-0)e crude product was purified by column chromatography (hexane/EtOAc = $5:1$) to provide the titled compound proximal-10Bh (33 mg, 65%) as a colorless solid, the regiochemistry of which was determined by NOESY spectrum. Mp 130−133 °C. ¹H NMR (500 MHz, CDCl₃) δ: 1.37 (12 H, s), 2.51 (3 H, s), 7.26−7.33 (10 H, m) 7.91 (1 H, d, J = 1.5 Hz), 7.97 (1 H, s), 8.47 (1 H, s). ¹³C NMR (125 MHz, CDCl₃) δ : 21.1, 24.8, 64.9, 84.5, 122.6, 127.6, 128.3, 128.6, 133.1, 135.0, 139.5, 140.0, 146.0. IR (CHCl₃): 2981 cm⁻¹. HRMS calcd for C₂₆H₂₉BN₃O₂ (M + H)⁺ m/z: 426.2353, found 426.2347.

1-(Adamantan-1-yl)-5-methyl-7-(4,4,5,5-tetramethyl-1,3,2 dioxaborolan-2-yl)-1H-benzo[d]triazole (proximal-10Bj) (Table 3, entry 8). Following the general procedure I, a mixture of 14B (61 mg, 101 μ mol), 1-azidoadamantane $6j^{10c}$ (54 mg, 0.30 mmol), and *i*-PrMgCl·LiCl (1.3 M in THF, 95 μL, 124 μmol) in Et₂O (1.0 mL, 0.10 [M](#page-3-0)) was stirred for 30 min at 0 °C. Th[e cr](#page-17-0)ude product was purified by column chromatography (hexane/EtOAc = $5:1$) to provide the titled compound proximal-10Bj (14 mg, 34%) as a colorless solid, the regiochemistry of which was determined by NOESY spectrum. Mp 239−241 °C. ¹ H NMR (500 MHz, CDCl3) δ: 1.41 (12 H, s), 1.80− 1.95 (6 H, m), 2.29 (3 H, brs), 2.48 (3 H, s), 2.57 (6 H, brs), 7.55 (1 H, d, $J = 1.0$ Hz), 7.88(1 H, brs). ¹³C NMR (125 MHz, CDCl₃) δ : 21.0, 24.7, 29.8, 35.8, 41.6, 62.1, 84.5, 121.6, 131.4, 132.4, 136.0, 147.3. IR (CHCl₃): 2914 cm⁻¹. HRMS calcd for $C_{23}H_{33}BN_3O_2 (M + H)^+ m$ / z: 394.2660, found 394.2680.

1-(Adamantan-1-yl)-6-methyl-4-(4,4,5,5-tetramethyl-1,3,2 dioxaborolan-2-yl)-1H-benzo[d]triazole (distal-10Bj) (Table 3, entry 8). This product, the regiochemistry of which was determined by NOESY spectrum, was obtained by the column chromatography (EtOAc) of the above-mentioned crude product and the followi[ng](#page-3-0) treatment with pinacol in CH_2Cl_2 to give a mixture of distal-10Bj $(34%)$, Et₂O, and pinacol $(1:0.072:0.065)$ as a colorless solid. Mp 174−180 °C.1 H NMR (500 MHz, CDCl3) δ: 1.42 (12 H, s), 1.84 (6 H, brs), 2.29 (3 H, brs), 2.46 (6 H, brs), 2.51 (3 H, s), 7.62 (1 H, s), 7.64 (1 H, s). ¹³C NMR (125 MHz, CDCl₃) δ: 21.8, 24.8, 29.5, 36.1, 42.1, 61.1, 84.1, 114.3, 131.4, 133.4, 135.4, 148.6. IR (CHCl₃): 2918 cm⁻¹. HRMS calcd for C₂₃H₃₂BN₃NaO₂ (M + Na)⁺ m/z: 416.2485, found 416.2485.

6-Methyl-4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)- 1H-indazole (proximal-10Bn) (Table 3, entry 9). An oven-dried flask was charged with $12B$ (43 mg, 88 μ mol). The flask was capped with a rubber septum, then evacuated, and backfilled with argon. Anhydrous Et_2O (0.90 mL) and (diazom[et](#page-3-0)hyl)trimethylsilane 6n (2.0 M in $Et₂O$ 0.14 mL, 0.28 mmol) were added, and the mixture was cooled to −78 °C. A solution of i-PrMgCl (2.0 M in THF, 0.060 mL,

0.12 mmol) was slowly added to the reaction mixture over 5 min. After stirring at −78 °C for 30 min, the reaction mixture was quenched with a saturated aqueous NH₄Cl solution. The reaction mixture was extracted with EtOAc, and the aqueous layer was extracted twice with EtOAc. The combined organic layers were washed with a saturated aqueous NaCl solution and dried over anhydrous $Na₂SO₄$. The solution was filtered through a glass filter and concentrated under reduced pressure. The crude product was purified by column chromatography (hexane/EtOAc = 2:1) to provide the titled compound proximal-10Bn (12 mg, 54%) as a colorless solid, the regiochemistry of which was determined by NOESY spectrum. Mp 155−157 °C. ¹ H NMR (500 MHz, CDCl3) δ: 1.40 (12 H, s), 2.48 (3 H, s), 7.37 (1 H, s), 7.52 (1 H, s), 8.42 (1 H, s). 13C NMR (125 MHz, CDCl3) δ: 21.7, 24.9, 83.9, 112.1, 131.2, 136.3, 136.5, 139.9. IR (CHCl₃): 3472 cm⁻¹. HRMS calcd for C₁₄H₂₀BN₂O₂ (M + H)⁺ m/z: 259.1618, found 259.1631.

2-tert-Butyl-5-methyl-3-phenyl-7-(4,4,5,5-tetramethyl-1,3,2 dioxaborolan-2-yl)-2,3-dihydrobenzo[d]isoxazole (distal-10Br) **(Table 3, entry 10).** An oven-dried flask was charged with $12B^{14b}$ (56) mg, 0.11 mmol). The flask was capped with a rubber septum, then evacuated, and backfilled with argon. Anhydrous $Et₂O$ (1.2 m[L\)](#page-17-0) and N-tert-[bu](#page-3-0)tyl- α -phenylnitrone 6r (61 mg, 0.35 mmol) were added, and the mixture was cooled to −78 °C. A solution of i-PrMgCl·LiCl (1.3 M in THF, 0.18 mL, 0.23 mmol) was slowly added to the reaction mixture over 5 min. After stirring at −78 °C for 30 min, the reaction mixture was quenched with a saturated aqueous $NH₄Cl$ solution. The reaction mixture was extracted with EtOAc, and the aqueous layer was extracted twice with EtOAc. The combined organic layers were washed with a saturated aqueous NaCl solution and dried over anhydrous $Na₂SO₄$. The solution was filtered through a glass filter and concentrated under reduced pressure. The crude product was purified by column chromatography (hexane/EtOAc = $11:1$) to provide the titled compound distal-10Br (29 mg, 64%) as a colorless solid, the regiochemistry of which was determined by NOESY spectrum. Mp 104−107 °C. ¹ H NMR (500 MHz, CDCl3) δ: 1.07 (6 H, s), 1.13 (6 H, s), 1.19 (9 H, s), 2.34 (3 H, s), 5.78 (1 H, s), 6.78 (1 H, s), 7.12− 7.13 (4 H, m), 7.19 (2 H, d, J = 7.5 Hz). 13C NMR (125 MHz, CDCl3) δ: 21.3, 24.5, 24.6, 25.5, 61.3, 66.8, 83.6, 109.8, 126.8, 128.0, 128.27, 128.34, 132.4, 138.3, 144.6, 157.6. IR (CHCl₃): 1582 cm⁻¹. . HRMS calcd for $C_{24}H_{33}BNO_3$ $(M + H)^+$ m/z : 394.2553, found 394.2546.

1-(2,6-Dibromophenyl)-5-methyl-7-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-1H-benzo[d]triazole (proximal-10Bd) (Table 3, entry 11). Following the general procedure I, a mixture of 14B (50 mg, 0.080 mmol), 2-azido-1,3-dibromobenzene $6d^{39}$ (69 mg, 0.25 mmol), and i-PrMgCl·LiCl (1.3 M in THF, 0.080 mL, 0.11 mmol) [in](#page-3-0) Et₂O (0.80 mL) was stirred for 30 min at 0 $^{\circ}$ C. The [c](#page-17-0)rude product was purified by column chromatography (hexane/ $EtOAc =$ 4:1) to provide the titled compound *proximal*-10Bd (28 mg, 67%) as a colorless solid. Mp 181−184 °C. ¹H NMR (500 MHz, CDCl₃) δ: 1.07 $(12 \text{ H}, \text{s})$, 2.55 $(3 \text{ H}, \text{s})$, 7.30 $(1 \text{ H}, \text{t}, J = 8.0 \text{ Hz})$, 7.68 $(2 \text{ H}, \text{d}, J = 8.0 \text{ Hz})$ Hz), 7.92 (1 H, d, $J = 1.0$ Hz), 8.04 (1 H, brs). ¹³C NMR (125 MHz, CDCl3) δ: 21.1, 24.4, 83.5, 122.7, 125.2, 131.6, 131.7, 133.7, 134.8, 137.9, 139.7, 145.9. IR (CHCl₃): 1478 cm⁻¹. HRMS calcd for $C_{19}H_{20}BN_3^{79}Br^{81}BrNaO_2 (M + Na)^+ m/z$: 515.9893, found 515.9882.

1-Cinnamyl-5-methyl-7-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-1H-benzo[d]triazole (proximal-10Bi) (Table 3, entry 12). Following the general procedure I, a mixture of 14B (55 mg, 90 μ mol), (E)-(3-azidoprop-1-en-1-yl)benzene $6i^{10c}$ (43 mg, 0.27 mmol), and *i*-PrMgCl·LiCl (1.3 M [i](#page-3-0)n THF, 0.090 mL, 110 μ mol) in Et₂O (0.90 mL, 0.10 M) was stirred for 30 min at 0 [°](#page-17-0)C. The crude product was purified by column chromatography (hexane/EtOAc = 40:9) to provide the titled compound proximal-10Bi (27 mg, 80%) as a colorless solid, the regiochemistry of which was determined by NOESY spectrum. Mp 134−136 °C. ¹H NMR (500 MHz, CDCl₃) *δ*: 1.40 (12 H, s), 2.52 (3 H, s), 5.86 (2 H, dd, $J = 1.0$, 6.0 Hz), 6.35 (1 H, d, J = 16 Hz), 6.43 (1 H, dt, J = 6.5, 16 Hz), 7.18–7.27 (5 H, m), 7.88 $(1 H, d, J = 1.5 Hz)$, 7.95 $(1 H, brs)$. ¹³C NMR $(125 MHz, CDCl₃)$ δ : 21.1, 24.9, 50.9, 84.4, 122.5, 124.6, 126.3, 127.7, 128.5, 132.0, 133.1,

134.0, 136.2, 139.2, 146.6. IR (CHCl₃): 3421 cm⁻¹. HRMS calcd for $C_{22}H_{27}BN_3O_2 (M + H)^+ m/z$: 376.2196, found 376.2193.

1-(2-Iodobenzyl)-5-methyl-7-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-1H-benzo[d]triazole (proximal-10Bk) (Table 3, entry 13). Following the general procedure I, a mixture of 14B (58 mg, 97 μ mol), 1-(azidomethyl)-2-iodobenzene 6 k^{10c} (76 mg, 0.29 mmol), and i -PrMgCl·LiCl (1.3 M in THF, 0.090 mL, 117 μ mol) in [E](#page-3-0)t₂O (1.0 mL, [0](#page-17-0).10 M) was stirred for 30 min at 0 °C. The crude product was purified by column chromatography (hexane/EtOAc $=$ 5:1) to provide the titled compound proximal-10Bk (33 mg, 71%) as a colorless solid. Mp 185−187 °C. ¹H NMR (500 MHz, CDCl₃) δ: 1.13 $(12 \text{ H}, \text{s})$, 2.54 $(3 \text{ H}, \text{s})$, 5.93 $(1 \text{ H}, \text{ dd}, J = 1.0, 7.5 \text{ Hz})$, 6.22 $(2 \text{ H}, \text{s})$, 6.91 (1 H, dt, $J = 1.0$, 7.5 Hz), 7.05 (1 H, t, $J = 7.5$ Hz), 7.88 (1 H, d, J $= 7.5$ Hz), 7.90 (1 H, d, J = 1.0 Hz), 8.0 (1 H, brs). ¹³C NMR (125 MHz, CDCl3) δ: 21.1, 24.4, 58.4, 84.3, 95.9, 122.7, 125.7, 128.4, 128.6, 133.4, 134.8, 139.1, 139.7, 146.7. IR (CHCl₃): 2982 cm⁻¹. HRMS calcd for $C_{20}H_{24}BIN_3O_2 (M + H)^+ m/z$: 476.1006, found 476.1017.

Methyl 2-[(5-Methyl-7-(4,4,5,5-tetramethyl-1,3,2-dioxaboro-10BI)^{(Table 3, entry 14). Following the general procedure I, a} mixture of 14B (57 mg, 95 μmol), ethyl methyl 2-(azidomethyl) benzoate $6I^{40}$ (56 mg, 0.29 mmol), and *i*-PrMgCl·LiCl (1.3 M in THF, 0.090 mL, 117 μ mol) in Et₂O (1.0 mL, 0.10 M) was stirred for 30 min at 0 °C. T[he](#page-17-0) crude product was purified by column chromatography $(hexane/EtOAC = 4:1)$ to provide the titled compound *proximal*-10Bl (29 mg, 74%) as a colorless solid. Mp 127−129 °C. ¹ H NMR (500 MHz, CDCl₃) δ : 1.05 (12 H, s), 2.53 (3 H, s), 3.97 (3 H, s), 6.01 (1 H, d, $J = 7.5$ Hz), 6.73 (2 H, s), 7.21 (1 H, dt, $J = 1.0$, 7.5 Hz), 7.27 (1 H, dt, $J = 1.0, 7.5$ Hz), 7.86 (1 H, d, $J = 1.0$ Hz) 8.0 (1 H, brs) 8.11 (1 H, dd, J = 1.0, 7.5 Hz). ¹³C NMR (125 MHz, CDCl₃) δ : 21.1, 24.3, 52.0, 52.7, 84.1, 122.6, 125.4, 126.7, 126.8, 131.1, 132.9, 133.3, 134.9, 139.3, 140.5, 146.7, 167.1. IR (CHCl₃): 1717 cm⁻¹. HRMS calcd for $C_{22}H_{27}BN_3O_4$ $(M + H)^+$ m/z : 408.2095, found 408.2100.

2-[(5-Methyl-7-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2 yl)-1H-benzo[d]triazol-1-yl)methyl]benzonitrile (proximal-
10Bm) (Table 3, entry 15). Following the general procedure I, a mixture of 14B (62 mg, 102 μ mol), 2-(azidomethyl)benzonitrile 6m⁴¹ (49 mg, 0.31 mmol), and i-PrMgCl·LiCl (1.3 M in THF, 0.10 mL, 124 μ mol) in Et₂O [\(1.](#page-3-0)0 mL, 0.10 M) was stirred for 30 min at 0 °C. T[he](#page-17-0) crude product was purified by column chromatography (hexane/ EtOAc = 3:1) to provide the titled compound *proximal*-10Bm (30 mg, 78%) as a colorless solid. Mp 174−176 °C. ¹ H NMR (500 MHz, CDCl3) δ: 1.16 (12 H, s), 2.54 (3 H, s), 6.19−6.22 (1 H, m), 6.58 (2 H, s), 7.30–7.34 (2 H, m), 7.71–7.74 (1 H, m), 7.92 (1 H, d, J = 1.5 Hz), 8.01 (1 H, brs). ¹³C NMR (125 MHz, CDCl₃) δ : 21.1, 24.4, 51.3, 84.4, 109.8, 117.1, 122.9, 125.5, 127.5, 132.7, 133.2, 133.7, 134.7, 140.0, 141.7, 146.7. IR (CHCl₃): 2226 cm⁻¹. HRMS calcd for $C_{21}H_{24}BN_4O_2$ $(M + H)^+$ m/z: 375.1992, found 375.1984.

General Procedure II: Distal-Selective (3 + 2) Cycloadditions of Silylbenzynes 5 (Tables 1 and 3). An oven-dried flask was charged with a silylbenzyne precursor 16 (1.0 equiv), capped with a rubber septum, and evacuated, and backfilled with argon. Anhydrous THF (0.10 M) was added via a [sy](#page-1-0)ringe, [a](#page-3-0)nd the reaction mixture was cooled to 0 °C. After 5 min, a dipole 6 (3.0 equiv) and Bu₄NF (1.0 M THF solution, 2.0 equiv) were added in that order, and the reaction mixture was stirred at 0 °C for 30 min. The mixture was filtered through a short pad of silica gel (hexane/EtOAc = 10:1 as the eluent). The effluent mixture was concentrated under reduced pressure and further purified by silica gel flash column chromatography to give silyl benzazole 11.

1-(4-Methoxyphenyl)-4-(trimethylsilyl)-1H-benzo[d]triazole (distal-11Aa) (Table 1, entry 5). Following the general procedure II, a mixture of $16A^{17}$ (100 mg, 0.27 mmol), 1-azido-4-methoxybenzene $6a^{34}$ (124 mg, 0.83 mmol), and Bu₄NF (1.0 M in THF, 0.54 mL, 0.54 mmol) in THF ([2.7](#page-17-0) [mL](#page-1-0), 0.10 M) was stirred for 30 min at 0 °C. The ra[tio](#page-17-0) of *distal-* to *proximal-11Aa* (5.7:1) was determined using ¹H NMR spectra of the crude product. The crude product was purified by column chromatography (hexane/EtOAc = $10:1$ to $5:1$) to provide the titled compound distal-11Aa (39 mg, 49%) as a colorless solid. Mp 97−100 °C. ¹H NMR (500 MHz, CDCl₃) δ: 0.53 (9 H, s), 3.90 (3 H, s), 7.11 (2 H, dd, J = 2.0, 8.5 Hz), 7.47 (1 H, dd, J = 1.0, 7.0 Hz), 7.51

 $(1 H, dd, J = 7.0, 8.0 Hz), 7.65 (1 H, dd, J = 1.0, 8.0 Hz), 7.66 (2 H,$ dd, J = 2.0, 8.5 Hz). ¹³C NMR (125 MHz, CDCl₃) δ : −0.77, 55.6, 110.7, 114.9, 124.5, 127.3, 129.6, 130.2, 131.4, 133.9, 150.2, 159.6. IR (CHCl₃): 1520 cm⁻¹. HRMS calcd for C₁₆H₁₉N₃NaOSi (M + Na)⁺ m/z: 320.1195, found 320.1192.

1-(4-Methoxyphenyl)-7-(trimethylsilyl)-1H-benzo[d]triazole (proximal-11Aa) (Table 1, entry 5). Product (7.9 mg, 10%) was obtained by column chromatography of the above-mentioned crude product as a colorless solid. Mp 106−109 °C. ¹H NMR (500 MHz, CDCl₃) δ : 0.02 (9 H, s), 3.[92](#page-1-0) (3 H, s), 7.08 (2 H, dd, J = 2.0, 7.0 Hz), 7.39 (1 H, dd, J = 7.0, 7.5 Hz), 7.44 (2 H, dd, J = 2.0, 7.0 Hz), 7.70 (1 H, d, $J = 7.0$ Hz), 8.14 (1 H, d, $J = 7.0$ Hz). ¹³C NMR (125 MHz, CDCl3) δ: 0.51, 55.7, 114.5, 121.3, 122.6, 123.6, 129.3, 130.9, 136.1, 138.3, 145.0, 160.9. IR (CHCl₃): 1518 cm⁻¹. HRMS calcd for $C_{16}H_{19}N_3NaOSi (M + Na)^+ m/z$: 320.1195, found 320.1197.

1-(4-(tert-Butyl)phenyl)-6-methyl-4-(trimethylsilyl)-1Hbenzo[d]triazole (distal-11Bb) (Table 3, entry 1). Following the general procedure II, a mixture of $16B^{14a}$ (70 mg, 0.20 mmol), 1azido-4- $(t$ ert-butyl)benzene $6b^{35}$ (105 mg, 0.60 mmol), and Bu₄NF (1.0 M in THF, 0.40 mL, 0.40 mmol) in [TH](#page-17-0)F (2.0 mL) was stirred for 1.0 h. The ratio of distal- to [pro](#page-17-0)ximal-11Bb (10:1) was determined using ¹H NMR spectra of the crude product. The crude product was purified by column chromatography (hexane/EtOAc = 30:1) to provide the titled compound distal-11Bb (42 mg, 61%) as a yellow solid. Mp 92−94 °C. ¹H NMR (500 MHz, CDCl₃) δ: 0.52 (9 H, s), 1.40 (9 H, s), 2.53 (3 H, s), 7.32 (1 H, s), 7.49 (1 H,s), 7.61 (2 H, d, J $= 8.0$ Hz), 7.68 (2 H, d, J = 8.0 Hz). ¹³C NMR (125 MHz, CDCl₃) δ : −0.7, 22.0, 31.3, 34.8, 110.2, 122.6, 126.6, 131.7, 131.9, 133.2, 134.7, 137.7, 149.0, 151.5. IR (CHCl₃): 2964, 1600, 1519, 1411 cm⁻¹. HRMS calcd for $C_{20}H_{28}N_3Si (M + H)^+ m/z$: 338.2053, found 338.2055.

1-(4-(tert-Butyl)phenyl)-5-methyl-7-(trimethylsilyl)-1Hbenzo[d]triazole (proxiaml-11Bb) (Table 3, entry 1). Product (12 mg, 18%) was obtained by column chromatography of the abovementioned crude product as a colorless oil. ¹H NMR (500 MHz, CDCl_{[3](#page-3-0)}) δ : −0.02 [\(](#page-3-0)9 H, s), 1.40 (9 H, s), 2.53 (3 H, s), 7.43 (2 H, d, J $= 8.0$ Hz), 7.51 (1 H, s), 7.58 (2 H, d, J = 8.0 Hz), 7.89 (1 H, s). ¹³C NMR (125 MHz, CD₃OD) δ: 0.4, 21.4, 31.3, 34.9, 120.1, 122.2, 126.3, 127.4, 133.3, 135.5, 136.7, 138.4, 145.8, 153.7. IR (CHCl₃): 1602, 1514, 1261, 1236 cm⁻¹. HRMS calcd for C₂₀H₂₈N₃Si (M + H)⁺ m/z: 338/.2053, found 338.2059.

1-(2,5-Dimethoxyphenyl)-6-methyl-4-(trimethylsilyl)-1Hbenzo[d]triazole (distal-11Bc) (Table 3, entry 2). Following the general procedure II, a mixture of $16B^{14a}$ (50 mg, 0.13 mmol), 2azido-1,4-dimethoxybenzene $6c^{10c}$ (69 mg, 0.39 mmol), and Bu₄NF (1.0 M in THF, 0.26 mL, 0.26 mmol) in [TH](#page-17-0)F (1.3 mL) was stirred for 0.5 h. The ratio of distal- to [prox](#page-17-0)imal-11Bc (14:1) was determined using ¹H NMR spectra of the crude product. The crude product was purified by column chromatography (hexane/EtOAc = $15:1$) to provide the titled compound distal-11Bc (30 mg, 68%) as a colorless solid, the regiochemistry of which was determined by NOESY spectrum. Mp 140−142 °C. ¹H NMR (500 MHz, CDCl₃) δ: 0.52 (9 H, s), 2.50 (3 H, s), 3.74 (3 H, s), 3.81 (3 H, s), 7.04 (1 H, d, $J = 2.0$ Hz), 7.07 (1 H, dd, J = 2.0, 8.0 Hz), 7.09 (1 H, d, J = 8.0 Hz), 7.13 (1 H, brs), 7.30 (1 H, d, J = 2.0 Hz). ¹³C NMR (125 MHz, CDCl₃) δ : −0.7, 22.0, 55.9, 56.5, 111.0, 113.4, 113.8, 116.2, 126.0, 131.65, 132.8, 133.3, 137.2, 147.7, 148.2, 153.8. IR (CHCl₃): 2959, 1595, 1514 cm⁻¹. . HRMS calcd for $C_{18}H_{24}N_3O_2Si$ $(M + H)^+$ m/z : 342.1638, found 342.1622.

1-(2,5-Dimethoxyphenyl)-5-methyl-7-(trimethylsilyl)-1Hbenzo[d]triazole (proximal-11Bc) (Table 3, entry 2). Product (2.7 mg, 6.0%) was obtained by column chromatography of the abovementioned crude product as a colorless solid. Mp 99−100 °C. ¹ H NMR (500 MHz, CDCl₃) δ: 0.01 (9 H, s), 2.[53](#page-3-0) (3 H, s), 3.64 (3 H, s), 3.81 (3 H, s), 6.99 (1 H, d, $J = 8.0$ Hz), 7.02 (1 H, d, $J = 2.0$ Hz), 7.11 (1 H, dd, $J = 2.0$, 8.0 Hz), 7.49 (1 H, s), 7.90 (1 H, s). ¹³C NMR (125 MHz, CD₃OD) δ: 1.2, 22.1, 57.2, 57.3, 115.2, 118.4, 119.3, 121.1, 124.6, 128.5, 136.1, 139.5, 140.4, 147.1, 152.6, 155.6. IR (CHCl₃): 2960, 1620, 1514 cm⁻¹. HRMS calcd for $C_{18}H_{24}N_3O_2Si (M + H)^+ m/$ z: 342.1638, found 342.1615.

1-Benzyl-4-(trimethylsilyl)-1H-benzo[d]triazole (distal-11Ae) (Table 3, entry 3). Following the general procedure II, a mixture of $16A^{14a}$ (50 mg, 0.14 mmol), (azidomethyl)benzene 6e (54 µL, 0.42 mmol), and Bu4NF (1.0 M in THF, 0.28 mL, 0.28 mmol) in THF (1.4 mL[\) wa](#page-17-0)[s](#page-3-0) stirred for 0.5 h. The ratio of distal- to proximal-11Ae (3.6:1) was determined using ¹H NMR spectra of the crude product. The crude product was purified by column chromatography (hexane/ EtOAc = 10:1) to provide the titled compound *distal*-11Ae (14.9 mg, 38%) as a colorless oil. ¹H NMR (500 MHz, CDCl₃) δ : 0.49 (9 H, s), 5.83 (2 H, s), 7.29–7.36 (7 H, m), 7.43 (1 H, dd, J = 2.0, 8.0 Hz). ¹³C NMR (125 MHz, CDCl₃) δ: −0.8, 52.1, 110.2, 126.6, 127.6, 128.3, 128.9, 129.3, 131.5, 133.8, 134.9, 150.4. IR (CHCl₃): 1724, 1597, 1497 cm⁻¹. HRMS calcd for C₁₆H₂₀N₃Si (M + H)⁺ m/z: 282.1427, found 282.1454.

1-Benzyl-7-(trimethylsilyl)-1H-benzo[d]triazole (proximal-**11Ae) (Table 3, entry 3).** Product $(5.3 \text{ mg}, 14\%)$ was obtained by column chromatography of the above-mentioned crude product as a colorless oil. ¹[H N](#page-3-0)MR (500 MHz, CDCl₃) δ : 0.35 (9 H, s), 6.05 (2 H, s), 6.89 (2 H, d, J = 8.0 Hz), 7.27–7.30 (3 H, m), 7.38 (1 H, t, J = 8.0) Hz), 7.68 (1 H, d, J = 8.0 Hz), 8.15 (1 H, dd, J = 2.0, 8.0 Hz). ¹³C NMR (125 MHz, CD₃OD) δ: 1.5, 50.1, 122.6, 124.3, 126.2, 127.9, 129.7, 130.7, 137.8, 138.8, 139.6, 146.8. IR (CHCl₃): 2396, 1778, 1620, 1527, 1497 cm⁻¹. HRMS calcd for C₁₆H₂₀N₃Si (M + H)⁺ m/z: 282.1427, found 282.1446.

1-Benzyl-6-methyl-4-(trimethylsilyl)-1H-benzo[d]triazole (distal-11Be) (Table 3, entry 4). Following the general procedure II, a mixture of $16B^{14a}$ (50 mg, 0.13 mmol), (azidomethyl)benzene 6e $(50 \,\mu L, 0.39 \text{ mmol})$, and Bu₄NF $(1.0 \text{ M in THF}, 0.26 \text{ mL}, 0.26 \text{ mmol})$ in THF (1.3 mL) [wa](#page-17-0)s [st](#page-3-0)irred for 0.5 h. The ratio of distal- to proximal-11Be $(3.3:1)$ was determined using ${}^{1}H$ NMR spectra of the crude product. The crude product was purified by column chromatography (hexane/EtOAc = $10:1$) to provide the titled compound *distal*-11Be (22 mg, 57%) as a colorless oil, the regiochemistry of which was determined by NOESY spectrum. ¹H NMR (500 MHz, CDCl₃) δ : 0.64 (9 H, s), 2.60 (3 H, s), 5.92 (2 H, s), 7.40−7.49 (7 H, m). 13C NMR (125 MHz, CDCl₃) δ: −0.8, 21.9, 51.8, 109.3, 127.4, 128.1, 128.8, 131.6, 132.0, 133.0, 135.1, 136.9, 148.9. IR (CHCl₃): 2277, 1605 cm⁻¹. HRMS calcd for C₁₇H₂₂N₃Si (M + H)⁺ m/z: 296.1583, found 296.1554.

1-Benzyl-5-methyl-7-(trimethylsilyl)-1H-benzo[d]triazole (*proximal-11Be*) (Table 3, entry 4). Product $(6.5 \text{ mg}, 18\%)$ was obtained by column chromatography of the above-mentioned crude product as a yellow oil. ¹H NMR (500 MHz, CDCl₃) δ : 0.34 (9 H, s), 2.52 (3 H, s), 6.02 (2 H, s)[, 6](#page-3-0).87 (2 H, d, J = 8.0 Hz), 7.26−7.28 (3 H, m), 7.48 (1 H, brs), 7.90 (1 H, brs). ¹³C NMR (125 MHz, CDCl₃) δ : 0.7, 21.4, 52.4, 120.5, 121.1, 126.1, 127.8, 128.8, 133.1, 136.1, 136.3, 137.6, 145.9. IR (CHCl₃): 2463, 1620, cm⁻¹. HRMS calcd for $C_{17}H_{22}N_3Si$ $(M + H)^+$ m/z: 296.1583, found 296.1605.

1-Benzyl-4-(tert-butyldimethylsilyl)-6-methyl-1H-benzo[d] triazole (distal-11Be′) (Table 3, entry 4). Following the general procedure II, a mixture of 16B′ (50 mg, 0.12 mmol), (azidomethyl) benzene 6e (46 μ L, 0.36 mmol) and Bu₄NF (1.0 M in THF, 0.24 mL, 0.24 mmol) in THF (1.2 mL) wa[s](#page-3-0) stirred for 0.5 h. The ratio of distalto proximal-11Be' (10:1) was determined using ¹H NMR spectra of the crude product. The crude product was purified by column chromatography (hexane/EtOAc = 10:1) to provide the titled compound distal-11Be' (27 mg, 69%) as a colorless oil. ¹H NMR $(500 \text{ MHz}, \text{CDCl}_3)$ δ : 0.52 (6 H, s), 0.95 (9 H, s), 2.46 (3 H, s), 5.77 (2 H, s), 7.12 (1 H, brs), 7.24 (1 H, d, J = 2.0 Hz), 7.28−7.34 (5 H, m). ¹³C NMR (125 MHz, CDCl₃) δ: −4.9, 17.2, 22.0, 26.9, 51.8, 109.3, 127.5, 128.2, 128.9, 130.9, 132.1, 133.2, 135.2, 136.7, 149.5. IR (CHCl₃): 2927, 1605, 1406 cm⁻¹. HRMS calcd for C₂₀H₂₈N₃Si (M + H)⁺ m/z : 338.2053, found 338.2066.

1-Benzyl-7-(tert-butyldimethylsilyl)-5-methyl-1H-benzo[d] triazole (proximal-11Be′) (Table 3, entry 4). Product (3.1 mg, 8.0%) was obtained by column chromatography of the abovementioned crude product as a colorless oil. ¹H NMR (500 MHz, CDCl₃) δ : 0.37 (6 H, [s\)](#page-3-0), 0.88 (9 H, s), 2.53 (3 H, s), 6.00 (2 H, s), 6.83 (2 H, dd, J = 2.0, 8.0 Hz), 7.24–7.27 (3 H, m), 7.52 (1 H, d, J = 2.0 Hz), 7.91 (1 H, brs). ¹³C NMR (125 MHz, CDCl₃) δ: −2.3, 17.9, 21.4, 27.0, 52.4, 110.4, 118.6, 120.6, 126.1, 127.6, 128.7, 132.8, 136.6, 139.6, 146.0. IR (CHCl₃): 2929, 1606, 1566, 1497 cm⁻¹. HRMS calcd for $C_{20}H_{28}N_3Si$ $(M + H)^+$ m/z : 338.2053, found 338.2038.

1-Decyl-6-methyl-4-(trimethylsilyl)-1H-benzo[d]triazole (distal-11Bf) (Table 3, entry 5). Following the general procedure II, a mixture of $16B^{14a}$ (50 mg, 0.13 mmol), 1-azidodecane $6f^{35}$ (71 mg, 0.39 mmol), and $Bu₄NF (1.0 M in THF, 0.26 mL, 0.26 mmol)$ in THF (1.3 mL) was [stir](#page-17-0)[re](#page-3-0)d for 0.5 h. The ratio of distal- to pro[xim](#page-17-0)al-11Bf $(4.0.1)$ was determined using ${}^{1}H$ NMR spectra of the crude product. The crude product was purified by column chromatography (hexane/ EtOAc = 20:1) to provide the titled compound *distal*-11Bf (24 mg) 53%) as a colorless solid, the regiochemistry of which was determined by NOESY spectrum. Mp 58–59 °C. ¹H NMR (500 MHz, CD₃OD) δ : 0.44 (9 H, s), 0.88 (3 H, t, J = 7.5 Hz), 1.24–1.33 (14 H, m), 1.98 $(2 H, quint, J = 7.5 Hz), 2.54 (3 H, s), 4.65 (2 H, t, J = 7.5 Hz), 7.33$ $(1 H, d, J = 2.0 Hz)$, 7.51 $(1 H, d, J = 2.0 Hz)$. ¹³C NMR (125 MHz, CD₃OD) δ: −0.8, 14.1, 22.0, 22.6, 26.8, 29.1, 29.2, 29.3, 29.5, 29.7, 31.8, 47.9, 109.1, 131.5, 132.2, 133.0, 136.6, 148.6. IR (CHCl₃): 2928, 2857, 2099, 1605, 1466 cm⁻¹. HRMS calcd for $\rm C_{20}H_{36}N_3Si~(M + H)^+$ m/z: 346.2679, found 346.2684.

1-Decyl-5-methyl-7-(trimethylsilyl)-1H-benzo[d]triazole (proximal-11Bf) (Table 3, entry 5). Product (9.0 mg, 20%) was obtained by column chromatography of the above-mentioned crude product as a yellow oil. ¹H NMR (500 MHz, CDCl₃) δ : 0.46 (9 H, s), 0.87 (3 H, t, $J = 7.5$ Hz), [1.](#page-3-0)25−1.43 (14 H, m), 2.06 (2 H, quint, $J =$ 7.5 Hz), 2.49 (3 H, s), 4.65 (2 H, t, $J = 7.5$ Hz), 7.43 (1 H, d, $J = 2.0$ Hz), 7.82 (1 H, brs). ¹³C NMR (125 MHz, CD₃OD) δ : 0.8, 14.1, 21.3, 22.7, 26.8, 29.3, 29.4, 29.5, 30.8, 31.6, 31.8, 49.6, 120.3, 121.0, 132.8, 135.4, 137.0, 145.7. IR (CHCl₃): 2928, 2857, 1717, 1601, 1466 cm⁻¹ . HRMS calcd for $C_{20}H_{36}N_3Si$ $(M + H)^+$ m/z : 346.2679, found 346.2649.

Ethyl 2-(6-Methyl-4-(trimethylsilyl)-1H-benzo[d]triazol-1-yl) acetate (distal-11Bg) (Table 3, entry 6). Following the general procedure II, a mixture of $16B^{14a}$ (50 mg, 0.13 mmol), ethyl 2azidoacetate $6g^{10c}$ (50 mg, 0.39 mmol) and Bu₄NF (1.0 M in THF, 0.26 mL, 0.26 mmol) in THF (1.[3](#page-3-0) [m](#page-17-0)L) was stirred for 0.5 h. The ratio of distal- to pr[oxim](#page-17-0)al-11Bg (3.3:1) was determined using ¹H NMR spectra of the crude product. The crude product was purified by column chromatography (hexane/EtOAc = $20:1$) to provide the titled compound distal-11Bg (22 mg, 58%) as a yellow solid. Mp 58−⁶⁰ °C. ¹ ¹H NMR (500 MHz, CDCl₃) δ: 0.48 (9 H, s), 1.29 (3 H, t, J = 7.5 Hz), 2.52 (3 H, s), 4.26 (2 H, q, J = 7.5 Hz), 5.35 (2 H, s), 7.20 (1 H, brs), 7.28 (1 H, d, J = 2.0 Hz). ¹³C NMR (125 MHz, CDCl₃) δ : −0.8, 14.1, 22.0, 48.9, 62.2, 108.9, 131.8, 132.7, 133.3, 137.5, 148.7, 166.7. IR (CHCl₃): 1755, 1605 cm⁻¹. HRMS calcd for C₁₄H₂₂N₃O₂Si (M + H)⁺ m/z: 292.1481, found 292.1507.

Ethyl 2-(5-Methyl-1H-benzo[d]triazol-1-yl)acetate (Desilylated Product of proximal-11Bg) (Table 3, entry 6). Product (5.3 mg, 20%) was obtained by column chromatography of the abovementioned crude product as a colorless solid. Mp 104−105 °C. ¹ H NMR (500 MHz, CDCl₃) δ: 1.26 (3 H, t, J = [7.5](#page-3-0) Hz), 2.53 (3 H, s), 4.25 (2 H, q, $J = 7.5$ Hz), 5.39 (2 H, s), 7.35 (2 H, brs), 7.84 (1 H, brs). ¹³C NMR (125 MHz, CDCl₃) δ: 14.1, 21.5, 49.1, 62.3, 108.7, 119.1, 130.1, 131.9, 134.2, 146.6, 166.4. IR (CHCl₃): 1755, 1591 cm⁻¹. . HRMS calcd for $C_{11}H_{14}N_3O_2$ $(M + H)^+$ m/z : 220.1086, found 220.1084.

1-Benzhydryl-6-methyl-4-(trimethylsilyl)-1H-benzo[d] triazole (distal-11Bh) (Table 3, entry 7). Following the general procedure II, a mixture of $16B^{14c}$ (50 mg, 0.13 mmol), (azidomethylene)dibenzene $6h^{39}$ (82 mg, 0.39 mmol), and Bu₄NF (1.0 M in THF, 0.26 mL, 0.26 m[mo](#page-3-0)l) i[n T](#page-17-0)HF (1.3 mL) was stirred for 1.0 h. The ratio of distal- to p[ro](#page-17-0)ximal-11Bh (5.0:1) was determined using ¹H NMR spectra of the crude product. The crude product was purified by column chromatography (hexane/EtOAc = 15:1) to provide the titled compound distal-11Bh (27 mg, 55%) as a colorless solid, the regiochemistry of which was determined by NOESY spectrum. Mp 138−139 °C. ¹H NMR (500 MHz, CDCl₃) δ: 0.47 (9 H, s), 2.40 (3 H, s), 6.92 (1 H, s), 7.23−7.26 (7 H, m), 7.31−7.36 (5 H, m). ¹³C NMR (125 MHz, CDCl₃) δ: −0.7, 22.0, 66.7, 110.1, 128.2, 128.4, 128.7, 131.7, 132.3, 133.1, 136.8, 138.1, 148.8. IR (CHCl₃):

1601, 1497, 1454 cm⁻¹. HRMS calcd for C₂₃H₂₆N₃Si (M + H)⁺ m/z: 372.1896, found 372.1909.

1-Benzhydryl-5-methyl-7-(trimethylsilyl)-1H-benzo[d] triazole (proximal-11Bh) (Table 3, entry 7). Product (5.1 mg) 10%) was obtained by column chromatography of the abovementioned crude product as a colorless oil, the regiochemistry of which was determined by NOESY [sp](#page-3-0)ectrum. ¹H NMR (500 MHz, CDCl3) δ: 0.39 (9 H, s), 2.51 (3 H, s), 4.11−4.25 (1 H, m), 7.15 (4 H, dd, J = 2.0, 8.0 Hz), 7.31 (5 H, m), 7.39 (1 H, s), 7.49 (1 H, s), 7.90 (1 H, s). ¹³C NMR (125 MHz, CDCl₃) δ: 1.0, 21.3, 65.8, 120.6, 121.0, 128.0, 128.4, 128.5, 128.8, 133.1, 136.4, 137.7, 139.5. IR $(CHCl₃)$: 1719, 1568, 1495, 1450 cm⁻¹. HRMS calcd for C₂₃H₂₆N₃Si (M + H)⁺ m/z: 372.1896, found 372.1903.

1-(3,5,7-Adamantan-1-yl)-6-methyl-4-(trimethylsilyl)-1Hbenzo[d]triazole (distal-11Bj) (Table 3, entry 8). Following the general procedure II, a mixture of $16B^{14c}$ (50 mg, 0.13 mmol), 1azidoadamantane $6j^{10c}$ (69 mg, 0.39 mmol), and $\overline{B}u_4NF$ (1.0 M in THF, 0.26 mL, 0.26 mmol) in THF (1[.3 m](#page-17-0)L) was stirred for 0.5 h. The crude product [was](#page-17-0) purified by column chromatography (hexane/ EtOAc = 30:1) to provide the titled compound *distal*-11Bj (25 mg, 57%) as a colorless solid, the regiochemistry of which was determined by NOESY spectrum. Mp 198−199 °C. ¹H NMR (500 MHz, CDCl₃) δ : 0.47 (9 H, s), 1.86 (6 H, brs), 2.32 (3 H, s), 2.50 (6 H, d, J = 2.0 Hz), 2.53 (3 H, s), 7.23 (1 H, brs), 7.53 (1 H, brs). 13C NMR (125 MHz, CDCl₃) δ: −0.8, 22.1, 29.6, 36.2, 41.9, 61.0, 112.3, 130.8, 131.0, 133.0, 135.5, 149.6. IR (CHCl₃): 2855, 1730, 1597, 1456 cm⁻¹. HRMS calcd for C₂₀H₃₀N₃Si $(M + H)^+$ m/z: 340.2209, found 340.2210.

5-Methyl-7-(trimethylsilyl)-1H-indazole (distal-11Bn) (Table **3, entry 9).** Following the general procedure II, a mixture of $16B^{14c}$ (50 mg, 0.13 mmol), (diazomethyl)trimethylsilane 6n (2.0 M in Et₂O, 0.20 mL, 0.39 mmol), and Bu4NF (1.0 M in THF, 0.26 mL, 0[.26](#page-17-0) [m](#page-3-0)mol) in THF (1.3 mL) was stirred for 1.0 h. The ratio of distal- to proximal-11Bn $(1.4.1)$ was determined using $^1\mathrm{H}$ NMR spectra of the crude product. The crude product was purified by column chromatography (hexane/EtOAc = 5:1) to provide the titled compound distal-11Bn (16 mg, 59%) as a colorless solid, the regiochemistry of which was determined by NOESY spectrum after N^I-methylation using MeI and *t-*BuOK. Mp 137–139 °C. ¹H NMR $(500 \text{ MHz}, \text{CDCl}_3)$ δ : 0.43 (9 H, s), 2.46 (3 H, s), 7.34 (1 H, d, J = 2.0) Hz), 7.54 (1 H, brs), 8.01 (1 H, s), 10.6 (NH, brs). 13C NMR (125 MHz, CDCl3) δ: −0.7, 21.3, 120.2, 120.9, 122.3, 130.0, 134.4, 135.0, 142.5. IR (CHCl₃): 3479, 1776, 1578, 1530 cm⁻¹. HRMS calcd for $C_{11}H_{17}N_2Si$ $(M + H)^+$ m/z : 205.1161, found 205.1140.

6-Methyl-4-(trimethylsilyl)-1H-indazole (proximal-11Bn) (Table 3, entry 9). Product (10 mg, 39%) was obtained by column chromatography of the above-mentioned crude product as a colorless solid. Mp 57–58 °C. ¹H NMR (500 MHz, CDCl₃) δ: 0.41 (9 H, s), 2.49 (3 [H](#page-3-0), s), 7.14 (1 H, brs), 7.28 (1 H, brs), 8.11 (1 H, brs), 10.0 (NH, brs). ¹³C NMR (125 MHz, CDCl₃) δ: −0.6, 21.9, 109.9, 124.9, 129.3, 133.4, 135.5, 136.2, 139.8. IR (CHCl₃): 3469, 1782, 1531 cm⁻¹. . HRMS calcd for $C_{11}H_{17}N_2Si$ $(M + H)^+$ m/z : 205.1161, found 205.1149.

5-Methyl-7-(tert-butyldimethylsilyl)-1H-indazole (distal-11Bn′) (Table 3, entry 9). Following the general procedure II, a mixture of 16B′ (85 mg, 0.20 mmol), (diazomethyl)trimethylsilane 6n $(2.0$ M in $\mathrm{Et}_2\mathrm{O},$ 0.30 mL, 0.60 mmol) and $\mathrm{Bu}_4\mathrm{NF}$ $(1.0$ M in THF, 0.40 mL, 0.40 mmol[\)](#page-3-0) [i](#page-3-0)n THF (2.0 mL) was stirred for 1 h. The ratio of *distal-* to *proximal-*11Bn' (3.3:1) was determined using ¹H NMR spectra of the crude product. The crude product was purified by column chromatography (hexane/EtOAc = $5:1$) to provide the titled compound distal-11Bn′ (38 mg, 76%) as a colorless solid. Mp 199− 201 °C. ¹H NMR (500 MHz, CDCl₃) δ: 0.45 (6 H, s), 0.93 (9 H, s), 2.47 (3 H, s), 7.30 (1 H, s), 7.54 (1 H, s), 7.99 (1 H, s). ¹³C NMR (125 MHz, CDCl₃) δ : −5.2, 17.6, 21.3, 26.5, 117.9, 120.9, 122.4, 129.6, 134.0, 136.5, 143.1. IR (CHCl₃): 3483, 1722, 1600, 1463, 1257 cm⁻¹. HRMS calcd for C₁₄H₂₃N₂Si (M + H)⁺ m/z: 247.1631, found 247.1645.

6-Methyl-4-(tert-butyldimethylsilyl)-1H-indazole (proximal-11Bn′) (Table 3, entry 9). Product (12 mg, 24%) was obtained by column chromatography of the above-mentioned crude product as a

colorless solid. Mp 145−147 °C. ¹H NMR (500 MHz, CDCl₃) δ: 0.43 (9 H, s), 0.92 (9 H, s), 2.50 (3 H, s), 7.13 (1 H,s), 7.28 (1 H, s), 8.09 $(1 H, s)$. ¹³C NMR (125 MHz, CDCl₃) δ : −4.8, 17.3, 22.0, 26.7, 109.9, 125.8, 130.9, 131.1, 135.9, 136.5, 139.7. IR (CHCl₃): 3469, 1602, 1251 cm⁻¹. HRMS calcd for C₁₄H₂₃N₂Si (M + H)⁺ m/z: 247.1631, found 247.1652.

2-(tert-Butyl)-5-methyl-3-phenyl-7-(trimethylsilyl)-2,3 dihydrobenzo[d]isoxazole (distal-11Br) (Table 3, entry 10). Following the general procedure II, a mixture of $16B^{14c}$ (50 mg, 0.13 mmol), N-tert-butyl- α -phenylnitrone 6r (69 mg, 0.39 mmol), [and](#page-17-0) Bu4NF (1.0 M in THF, 0.26 mL, 0.26 mmol) in T[H](#page-3-0)[F](#page-17-0) (1.3 mL) was stirred for 3.0 h. The crude product was purified by column chromatography (hexane/EtOAc = 10:1) to provide the titled compound *distal-*11Br (41 mg, 94%) as a colorless oil. ¹H NMR $(500 \text{ MHz}, \text{CDCl}_3)$ δ: 0.32 (9 H, s), 1.17 (9 H, s), 2.20 (3 H, s), 5.51 $(1 H, s)$, 6.70 $(1 H, s)$, 7.00 $(1 H, s)$, 7.24 $(1 H, t, J = 8.0 Hz)$, 7.32 $(2 H, t, J = 8.0 Hz)$ H, t, $J = 8.0$ Hz), 7.36 (2 H, d, $J = 8.0$ Hz) ¹³C NMR (125 MHz, CDCl₃) δ: -1.2, 20.7, 25.4, 60.9, 66.7, 116.7, 125.1, 127.2, 127.4, 128.2, 128.5, 129.4, 134.2, 144.1, 159.2. IR (CHCl₃): 2976, 1602, 1454, 1393 cm⁻¹. HRMS calcd for C₂₁H₃₀NOSi (M + H)⁺ m/z: 340.2097, found 340.2090.

1-Benzyl-6-fluoro-4-(trimethylsilyl)-1H-benzo[d]triazole (distal-11Ce) (Table 3, entry 16). Following the general procedure II, a mixture of $16C^{14c}$ (50 mg, 0.13 mmol), (azidomethyl)benzene 6e (50 μ L, 0.39 mmol), and Bu₄NF (1.0 M in THF, 0.26 mL, 0.26 mmol) in THF (1.3 mL) [wa](#page-17-0)[s](#page-3-0) stirred for 0.5 h. The ratio of distal- to proximal-11Ce $(2.8:1)$ was determined using ${}^{1}H$ NMR spectra of the crude product. The crude product was purified by column chromatography (hexane/EtOAc = $10:1$) to provide the titled compound *distal*-11Ce $(27 \text{ mg}, 56\%)$ as a colorless oil. ¹H NMR (500 MHz, CDCl₃) δ : 0.48 $(9 H, s)$, 5.77 $(2 H, s)$, 6.91 $(1 H, dd, J = 2.0, 8.0 Hz)$, 7.16 $(1 H, dd, J)$ $= 2.0, 8.0$ Hz), 7.28–7.36 (5 H, m). ¹³C NMR (125 MHz, CDCl₃) δ : −0.9, 52.3, 95.6 (J = 25 Hz), 118.7 (J = 25 Hz), 127.7, 128.5, 129.0, 132.2 ($J = 10$ Hz), 134.5, 136.8 ($J = 10$ Hz), 147.1, 161.7 ($J = 240$ Hz). IR (CHCl₃): 1724, 1604, 1501 cm⁻¹. HRMS calcd for C₁₆H₁₉FN₃Si $(M + H)^+$ m/z: 300.1332, found 300.1336.

1-Benzyl-5-fluoro-1H-benzo[d]triazole (Desilylated Product of *proximal*-11Ce) (Table 3, entry 16). Product $(8.2 \text{ mg}, 28%)$ was obtained by column chromatography of the above-mentioned crude product as a colorless solid. Mp 101−103 °C. ¹ H NMR (500 MHz, CDCl₃) δ : 5.84 (2 H, s), 7.[18](#page-3-0) (1 H, dt, J = 2.0, 8.0 Hz), 7.27–7.37 (6 H, m), 7.68 (1 H, dd, J = 2.0, 8.0 Hz). ¹³C NMR (125 MHz, CDCl₃) δ: 52.6, 104.6 ($J = 26$ Hz), 110.7 ($J = 10$ HZ), 117.4 ($J = 26$ Hz), 127.6, 128.7, 129.1, 129.8, 134.3, 146.6 $(J = 10 \text{ Hz})$, 159.7 $(J = 240 \text{ Hz})$ Hz). IR (CHCl₃): 2396, 1778, 1627, 1595, 1499 cm^{−1}. HRMS calcd for $C_{13}H_{11}FN_3$ $(M + H)^+$ m/z : 228.0937, found 228.0916.

1-Benzyl-6-chloro-4-(trimethylsilyl)-1H-benzo[d]triazole (distal-11De) (Table 3, entry 17). Following the general procedure II, a mixture of $16D^{14c}$ (50 mg, 0.12 mmol), (azidomethyl)benzene 6e $(46 \,\mu L, 0.36 \text{ mmol})$, and Bu₄NF $(1.0 \text{ M in THF}, 0.24 \text{ mL}, 0.24 \text{ mmol})$ in THF (1.2 mL) w[as](#page-17-0) [sti](#page-3-0)rred for 1.0 h. The ratio of distal- to proximal-11De $(3.0:1)$ was determined using ${}^{1}H$ NMR spectra of the crude product. The crude product was purified by column chromatography (hexane/EtOAc = $20:1$) to provide the titled compound *distal*-11De $(12 \text{ mg}, 31\%)$ as a colorless oil. ¹H NMR (500 MHz, CDCl₃) δ : 0.48 (9 H, s), 5.78 (2 H, s), 7.29−7.36 (7 H, m). 13C NMR (125 MHz, CDCl3) δ: −0.9, 52.2, 109.7, 127.6, 128.6, 129.1, 130.0, 132.3, 133.5, 134.4, 136.0, 148.8. IR (CHCl₃): 2959, 1750, 1593, 1497 cm⁻¹. HRMS calcd for $C_{16}H_{19}CN_3Si (M + H)^+ m/z$: 316.1037, found 316.1048.

1-Benzyl-5-chloro-1H-benzo[d]triazole (Desilylated Product of proximal-11De) (Table 3, entry 17). Product (3.0 mg, 12%) was obtained by column chromatography of the above-mentioned crude product as a colorless solid. Mp 148–149 °C. ¹H NMR (500 MHz, CDCl₃) δ : 5.84 (2 H, s), 7.[25](#page-3-0)−7.27 (4 H, m), 7.33−7.37 (3 H, m), 8.05 (1 H, d, J = 2.0 Hz). ¹³C NMR (125 MHz, CDCl₃) δ : 52.6, 110.7, 119.5, 127.69, 128.4, 128.7, 129.1, 129.9, 131.5, 134.3, 147.0. IR (CHCl₃): 1732, 1583, 1497, 1479 cm⁻¹. HRMS calcd for C₁₃H₁₁ClN₃ $(M + H)^+$ m/z: 244.0642, found 244.0624.

Ethyl 5-Methyl-7-(trimethylsilyl)-1H-indazole-3-carboxylate (distal-11Bp) (Table 3, entry 18). Following the general procedure II, a mixture of $16B^{14c}$ (50 mg, 0.13 mmol), ethyl 2-diazoacetate 6p $(40 \,\mu L, 0.39 \text{ mmol})$, and $Bu_4NF (1.0 \text{ M in THF}, 0.26 \text{ mL}, 0.26 \text{ mmol})$ in THF (1.3 mL) w[as s](#page-17-0)tirred for 0.5 h. The ratio of distal- to proximal-11Bp (12:1) was determined using ¹H NMR spectra of the crude product. The crude product was purified by column chromatography (hexane/EtOAc = $10:1$) to provide the titled compound *distal*-11Bp (30 mg, 83%) as a colorless solid, the regiochemistry of which was determined by NOESY spectrum. Mp 165−166 °C. ¹ H NMR (500 MHz, CDCl₃) δ: 0.44 (9 H, s), 1.48 (3 H, t, J = 8.0 Hz), 2.50 (3 H, s), 4.51 (2 H, q, J = 8.0 Hz), 7.36 (1 H, s), 8.03 (1 H, s), 10.35 (NH, brs). ¹³C NMR (125 MHz, CDCl₃) δ: −0.8, 14.4, 21.5, 60.9, 121.3, 121.9, 122.0, 132.6, 135.6, 136.1, 143.8, 162.7. IR (CHCl₃): 3466, 1715, 1601, 1452 cm⁻¹. HRMS calcd for C₁₄H₂₁N₂O₂Si (M + H)⁺ m/z: 277.1372, found 277.1351.

Ethyl 6-Methyl-4-(trimethylsilyl)-1H-indazole-3-carboxylate (proximal-11Bp) (Table 3, entry 18). Product $(3.6 \text{ mg}, 10\%)$ was obtained by column chromatography of the above-mentioned crude product as a colorless oil. ¹H NMR (500 MHz, CDCl₃) δ : 0.41 (9 H, s), 1.46 (3 H, t, J = 7.5 H[z\)](#page-3-0), 2.49 (3 H, s), 4.48 (2 H, q, J = 7.5 Hz), 7.34 (1 H, s), 7.39 (1 H, s). ¹³C NMR (125 MHz, CDCl₃) δ : 1.1, 14.5, 21.8, 61.2, 110.5, 128.8, 130.1, 133.8, 134.4, 136.6, 141.6, 163.1. IR (CHCl3): 3451, 1724, 1609, 1443 cm[−]¹ . HRMS calcd for $C_{14}H_{21}N_2O_2Si$ $(M + H)^+$ m/z : 277.1372, found 277.1356.

5-Methyl-3,3-diphenyl-7-(trimethylsilyl)-3H-indazole (distal-11Bq) (Table 3, entry 19). Following the general procedure II, a mixture of 16B14c (50 mg, 0.13 mmol), (diazomethylene)dibenzene $6q^{42}$ (76 μ L, 0.39 mmol), and Bu₄NF (1.0 M in THF, 0.26 mL, 0.26 mmol) in THF [\(1](#page-17-0).3 mL) was stirred for 1.0 h. The ratio of distal- to pr[oxi](#page-17-0)mal-11 Bq $(3.0.1)$ was determined using ¹H NMR spectra of the crude product. The crude product was purified by column chromatography (hexane/EtOAc = 50:1) to provide the titled compound distal-11 Bq (37 mg, 80%) as a colorless solid. Mp 122− 123 °C. ¹H NMR (500 MHz, CDCl₃) δ: 0.49 (9 H, s), 2.45 (3 H, s), 7.27−7.30 (10 H, m), 7.37 (1 H, s), 7.40 (1 H, s). 13C NMR (125 MHz, CD₃OD) δ: 0.6, 22.5, 102.3, 127.6, 129.6, 130.0, 130.5, 137.3, 137.9, 140.3, 142.3, 145.7, 161.7. IR (CHCl₃): 2959, 1591, 1493, 1466 cm⁻¹. HRMS calcd for C₂₃H₂₅N₂Si (M + H)⁺ m/z: 357.1787, found 357.1766 .

5-Methyl-3-phenyl-7-(trimethylsilyl)benzo[d]isoxazole (distal-11Bs) (Table 3, entry 20). Following the general procedure II, a mixture of $16B^{14c}$ (50 mg, 0.13 mmol), N-hydroxybenzimidoyl chloride⁴³ (61 mg, 0.39 mmol), and Bu₄NF (1.0 M in THF, 0.65 mL, 0.65 mmol) in TH[F \(](#page-17-0)1.3 mL) was stirred for 24 h. The crude product was puri[fi](#page-18-0)ed by column chromatography (hexane/EtOAc = $20:1$) to provide the titled compound distal-11Bs (27 mg, 73%) as a colorless solid, the regiochemistry of which was determined by NOESY spectrum. Mp 159–160 °C. ¹H NMR (500 MHz, CDCl₃) δ: 0.45 (9 H, s), 2.51 (3 H, s), 7.45 (1 H, d, J = 2.0 Hz), 7.51−7.58 (3 H, m), 7.68 (1 H, brs), 7.95 (2 H, dd, J = 2.0, 8.0 Hz). 13C NMR (125 MHz, CDCl3) δ: −1.2, 21.3, 119.1, 121.8, 122.2, 128.0, 129.0, 129.4, 129.9, 133.1, 136.8, 156.8, 167.1. IR (CHCl₃): 3059, 2361, 1595 cm⁻¹. . HRMS calcd for $C_{17}H_{20}NOSi (M + H)^+ m/z$: 282.1314, found 282.1323.

Synthesis of Hippadine Analogue 18 (Scheme 1a). 1-[(6- Iodobenzo[d][1,3]dioxol-5-yl)methyl]-7-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-1H-benzo[d]triazole (proximal-10At). Following the general procedure I, a mixture of 14A ([0.3](#page-4-0)0 g, 0.51 mmol), 5-(azidomethyl)-6-iodobenzo $[d]$ 1,3-dioxole 6t⁴³ (0.46 g, 1.5 mmol), and i-PrMgCl·LiCl (1.3 M in THF, 0.47 mL, 0.61 mmol) in Et₂O (5.1 mL) was stirred for 30 min at 0 $^{\circ}$ C. The cru[de](#page-18-0) product was purified by column chromatography (hexane/EtOAc = $10:1$) to provide the titled compound proximal-10At (178 mg, 69%) as a colorless solid. Mp 166−168 °C. ¹H NMR (500 MHz, CDCl₃) δ : 1.21 (12 H, s), 5.51 (1 H, s), 5.85 (2 H, s), 6.16 (2 H, s), 7.30 (1 H, s), 7.41 $(1 H, t, J = 7.5 Hz)$, 8.08 $(1 H, d, J = 7.5 Hz)$, 8.24 $(1 H, d, J = 7.5$ Hz). ¹³C NMR (125 MHz, CDCl₃) δ: 24.5, 58.0, 83.7, 84.4, 101.6, 106.9, 118.5, 123.6, 123.7, 133.2, 136.2, 137.8, 145.9, 147.4, 148.6. IR (CHCl₃): 2982, 1042 cm⁻¹. HRMS calcd for $C_{20}H_{21}BIN_3NaO_4$ (M + Na)⁺ m/z: 528.0567, found 528.0568.

1-((6-Iodobenzo[d][1,3]dioxol-5-yl)methyl)-5-methyl-7- (4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-1H-benzo[d] triazole (proximal-10Bt). Following the general procedure I, a mixture of 14B (0.20 g, 0.33 mmol), 5-(azidomethyl)-6-iodobenzo- $\left[d\right]$ 1,3-dioxole $6t^{43}$ (0.30 g, 0.99 mmol), and *i*-PrMgCl·LiCl (1.3 M in THF, 0.31 mL, 0.39 mmol) in THF (1.0 mL) was stirred for 30 min at 0 °C. Th[e](#page-18-0) crude product was purified by column chromatography (hexane/EtOAc = 7:1) to provide the titled compound *proximal*-10Bt (115 mg, 67%) as a colorless solid. Mp 163−165 °C. ¹ H NMR (500 MHz, CDCl₃) δ : 1.21 (12 H, s), 2.53 (3 H, s), 5.49 (1 H, s), 5.85 (2 H, s), 6.13 (2 H, s), 7.30 (1 H, s), 7.90 (1 H, s), 7.99 (1 H, s). 13C NMR (125 MHz, CDCl₃) δ: 21.1, 24.5, 58.0, 83.7, 84.4, 101.6, 106.9, 118.5, 122.8, 133.3, 133.5, 134.7, 139.8, 146.6, 147.4, 148.6. IR (CHCl₃): 2982, 1042 cm⁻¹. HRMS calcd for C₂₁H₂₃BIN₃NaO₄ (M + Na)⁺ m/z : 542.0724, found 542.0724.

7H-[1,3]Dioxolo[4,5-j]triazolo[4,5,1-de]phenanthridine (17A). An oven-dried round-bottom flask was charged with proximal-10At (50 mg, 0.096 mmol), $PdCl_2(dppf) \cdot CH_2Cl_2$ (24 mg, 0.029 mmol), and $Ba(OH)_2·8H_2O$ (91 mg, 0.29 mmol). The flask was capped with a rubber septum, then evacuated, and backfilled with argon. Anhydrous dioxane (0.50 mL) was added via a syringe, and the mixture was stirred at 100 °C for 16 h. After the reaction, the mixture was filtered through a short pad of Celite, washed with CH_2Cl_2 , and evaporated under reduced pressure. The residue was washed with acetone to provide the titled compound 17A (10 mg, 40%) as a colorless solid. Mp >350 °C. ¹H NMR (500 MHz, $(CD_3)_2$ SO, 80 °C) δ : 6.02 (2 H, s), 6.12 (2 H, s), 7.05 (1 H, s), 7.34 (1 H, t, J = 7.5 Hz), 7.68 (1 H, s), 7.76 (2 H, d, J = 7.5 Hz). 13C NMR (125 MHz, $(CD_3)_2$ SO, 80 °C) δ: 50.5, 103.3, 104.8, 109.5, 118.4, 118.7, 121.3, 123.1, 126.8, 126.9, 132.3, 144.6, 149.4, 150.0. IR (CHCl₃): 3008, 1039 cm⁻¹. HRMS calcd for C₁₄H₁₀N₃O₂ (M + H)⁺ m/z: 252.0773, found 252.0796.

2-Methyl-7H-[1,3]dioxolo[4,5-j]triazolo[4,5,1-de] phenanthridine (17B). An oven-dried round-bottom flask was charged with proximal-10Bt (50 mg, 0.096 mmol), $PdCl₂(dppf)·CH₂Cl₂$ (24 mg, 0.029 mmol), and Ba(OH)₂·8H₂O (91) mg, 0.29 mmol). The flask was capped with a rubber septum, then evacuated, and backfilled with argon. Anhydrous dioxane (0.50 mL) was added via a syringe, and the mixture was stirred at 100 °C for 16 h. After the reaction, the mixture was filtered through a short pad of Celite, washed with CH_2Cl_2 , and evaporated under reduced pressure. The residue was washed with acetone to provide the titled compound 17B (15 mg, 59%) as a colorless solid. Mp 267-269 °C. ¹H NMR $(500 \text{ MHz}, \text{CDCl}_3)$ δ : 2.53 $(3 \text{ H}, \text{s})$, 5.92 $(2 \text{ H}, \text{s})$, 6.07 $(2 \text{ H}, \text{s})$, 6.79 (1 H, s), 7.32 (1 H, s), 7.35 (1 H, s), 7.55 (1 H, s). 13C NMR (125 MHz, CDCl₃) δ: 22.2, 49.4, 101.9, 103.1, 107.7, 116.6, 118.9, 119.1, 122.2, 124.5, 129.9, 135.6, 144.2, 148.3, 148.7. IR (CHCl₃): 3008, 1039 cm⁻¹. HRMS calcd for C₁₅H₁₂N₃O₂ (M + H)⁺ m/z: 266.0930, found 266.0921.

Preparation of 7H-[1,3]Dioxolo[4,5-j]triazolo[4,5,1-de] phenanthridin-7-one (18A) from proximal-10At. An oven-dried round-bottom flask was charged with proximal-10At (50 mg, 0.099 mmol), $PdCl_2(dppf) \cdot CH_2Cl_2$ (24 mg, 0.030 mmol), and Ba- $(OH)_2·8H_2O$ (94 mg, 0.30 mmol). The flask was capped with a rubber septum, then evacuated, and backfilled with argon. Anhydrous dioxane (0.50 mL) was added via a syringe and the mixture was stirred for 18 h at 100 °C. After the reaction, the mixture was filtered through a short pad of Celite, washed with CH_2Cl_2 , and evaporated under reduced pressure to give crude 17A. Then, activated $MnO₂$ (85 mg, 0.99 mmol) was added into the crude 17A. The flask was capped with a rubber septum, then evacuated, and backfilled with argon. Anhydrous $CH₂Cl₂$ (3.3 mL) was added via a syringe, and the mixture was stirred for 24 h at room temperature. After the reaction, the mixture was filtered through a short pad of Celite, washed with CH_2Cl_2 , and evaporated under reduced pressure. The crude product was washed with ethanol to provide the titled compound 18A (10 mg, 53% from proximal-10At) as a pale yellow solid. Mp >350 $^{\circ}$ C. 1 H NMR (500 MHz, (CD_3) ₂SO, 80 °C) δ : 6.32 (2 H, s), 7.77 (1 H, t, J = 7.5 Hz), 7.90 (1 H, s), 8.13 (1 H, s), 8.33 (1 H, d, J = 7.5 Hz), 8.54 (1 H, d, J = 7.5 Hz). ¹³C NMR (125 MHz, $(CD_3)_2$ SO, 80 °C) δ : 103.8, 108.5,

117.7, 121.3, 123.2, 124.9, 127.8, 129.2, 131.4, 144.2, 149.9, 154.2, 155.4. IR (CHCl₃): 3008, 1715, 1039 cm⁻¹. HRMS calcd for $C_{14}H_8N_3O_3$ $(M + H)^+$ m/z : 266.0566, found 266.0583.

Preparation of 2-Methyl-7H-[1,3]dioxolo[4,5-j]triazolo[4,5,1 de]phenanthridin-7-one (18B) from proximal-10Bt. An ovendried round-bottom flask was charged with proximal-10Bt (50 mg, 0.096 mmol), $PdCl₂(dppf)·CH₂Cl₂$ (24 mg, 0.030 mmol), and $Ba(OH)_2·8H_2O$ (91 mg, 0.30 mmol). The flask was capped with a rubber septum, then evacuated, and backfilled with argon. Anhydrous dioxane (0.50 mL) was added, and the mixture was stirred at 100 °C for 18 h. After the reaction, the mixture was filtered through a short pad of Celite, washed with CH_2Cl_2 , and evaporated under reduced pressure to give crude 17B. Then, activated $MnO₂$ (80 mg, 0.96 mmol) was added into the crude 17B. The flask was capped with a rubber septum, then evacuated, and backfilled with argon. Anhydrous CH_2Cl_2 (3.3 mL) was added, and the mixture was stirred for 24 h at room temperature. After the reaction, the mixture was filtered through a short pad of Celite, washed with CH_2Cl_2 , and evaporated under reduced pressure. The residue was washed with acetone to provide the titled compound 18B (16 mg, 61% from proximal-10Bt) as a pale yellow solid. Mp 256−258 °C. ¹H NMR (500 MHz, CDCl₃) δ: 2.69 (3 H, s), 6.22 (2 H, s), 7.60 (1 H, s), 7.93 (2 H, s), 7.98 (1 H, s), 8.06 (1 H, s) . ¹³C NMR $(125 \text{ MHz}, \text{CDCl}_3)$ δ : 22.1, 102.2, 102.9, 109.5, 116.7, 120.7, 123.6, 124.4, 127.7, 130.9, 137.4, 144.9, 149.6, 153.8, 155.2. IR (CHCl₃): 3008, 1703, 1039 cm⁻¹. HRMS calcd for $C_{15}H_{10}N_3O_3$ $(M + H)^+$ m/z: 280.0722, found 280.0727.

Formal Synthesis of Vorozole 21 (Scheme 1b). 4-(tert- Butyldimethylsilyl)-6-(1,3-dioxolan-2-yl)-1-methyl-1H-benzo- [d]triazole (distal-11Eu'). Following the general procedure II, a mixture of 16E′ (100 mg, 0.158 mmol), (azidomethyl)t[rim](#page-4-0)ethylsilane 6u (62 mg, 0.48 mmol) (for safety reasons, commercially available 6u was used as an equivalent of the highly explosive methyl azide), and Bu4NF (1.0 M in THF, 0.32 mL, 0.32 mmol) in THF (1.6 mL) was stirred for 30 min at 0 °C. The ratio of distal- to proximal-11Eu' $(15:1)$ was determined using ${}^{1}H$ NMR spectra of the crude product. The crude product was purified by column chromatography (hexane/ EtOAc = 3:1) to provide the titled compound distal- $11Eu'$ (43 mg, 84%) as a colorless solid. Mp 92−94 °C. ¹H NMR (500 MHz, CDCl₃) δ: 0.52 (6 H, s), 0.93 (9 H, s), 4.09−4.20 (4 H, m), 4.28 (3 H, s), 6.00 $(1 H, s)$, 7.54 $(1 H, d, J = 8.0 Hz)$, 7.68 $(1 H, d, J = 8.0 Hz)$. ¹³C NMR (125 MHz, CDCl3) δ: −5.0, 17.2, 26.9, 34.1, 65.4, 103.5, 107.5, 129.5, 131.8, 133.2, 136.2, 151.0. IR (CHCl₃): 1602, 1408, 1328, 1236, 1120 cm⁻¹. HRMS calcd for C₁₆H₂₆N₃O₂Si (M + H)⁺ m/z: 320.1794, found 320.1801.

6-(1,3-Dioxolan-2-yl)-1-methyl-1H-benzo[d]triazole (24). Bu₄NF (1.0 M in THF, 0.35 mL, 350 μ mol) was added into the THF solution $(0.70 \text{ mL}, 0.10 \text{ M})$ of distal-11E' $(23 \text{ mg}, 71 \mu \text{mol})$ and stirred for 48 h at 70 °C. The reaction mixture was cooled to room temperature and evaporated under reduced pressure. The crude product was purified by column chromatography (hexane/EtOAc = 3:1) to provide the titled compound 24 (14 mg, 94%) as a colorless solid. Mp 97–98 °C. ¹H NMR (500 MHz, CDCl₃) δ: 4.08–4.18 (4 H, m), 4.31 (3 H, s), 5.98 (1 H, s), 7.49 (1 H, d, J = 8.0 Hz), 7.69 (1 H, s), 8.05 (1 H, d, J = 8.0 Hz). ¹³C NMR (125 MHz, CDCl₃) δ : 34.3, 65.4, 103.2, 107.1, 120.0, 122.6, 133.3, 137.7, 146.4. IR (CHCl₃): 1600, 1193 cm⁻¹. HRMS calcd for C₁₀H₁₂N₃O₂ (M + H)⁺ m/z: 206.0930, found 206.0910.

1-Methyl-1H-benzo[d]triazole-6-carbaldehyde (20). TsOH·- $H₂O$ (3.4 mg, 20 μ mol) was added into the acetone solution (0.70 mL) of 24 (14 mg), and the mixture was stirred at room temperature for 3 h. After the reaction, a saturated aqueous $NaHCO₃$ was added into the reaction mixture, and the mixture was extracted with $Et₂O$. The aqueous layer was extracted twice with $Et₂O$. The combined organic layers were washed with a saturated aqueous NaCl solution. The organic layer was dried over anhydrous $Na₂SO₄$ and evaporated under reduced pressure. The crude product was purified by column chromatography (hexane/EtOAc = $3:1$) to provide the titled compound ²⁰ (10 mg, 96%) as a colorless solid. Mp 119−¹²² °C. ¹ ¹H NMR (500 MHz, CDCl₃) δ : 4.41 (3 H, s), 7.91 (1 H, d, J = 8.0) Hz), 8.11 (1 H, s), 8.19 (1 H, d, J = 8.0 Hz), 10.19 (1 H, s). ¹³C NMR (125 MHz, CDCl3) δ: 34.7, 112.4, 120.9, 124.0, 133.5, 135.1, 148.6, 191.3. IR (CHCl₃): 1703, 1616, 1602, 1236 cm⁻¹. HRMS calcd for $C_8H_8N_3O (M + H)^+ m/z$: 162.0667, found 162.0692.

Synthesis of iso-Vorozole 23 (Scheme 1c). 5-(1,3-Dioxolan-2-yl)-7-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-1- ((trimethylsilyl)methyl)-1H-benzo[d]triazole (proximal-10Cu). Following the general procedure I, a mixture [o](#page-4-0)f 14C (94 mg, 0.14 mmol), (azidomethyl)trimethylsilane 6u (0.21 mL, 1.41 mmol), and *i*-PrMgCl·LiCl $(1.3 M$ in THF, 0.21 mL, 0.27 mmol) in Et₂O $(1.4 mL)$ was stirred for 30 min at 0 °C. The crude product was purified by column chromatography (hexane/EtOAc = $5:2$) to provide the titled compound proximal-10Cu (33 mg, 57%) as a colorless oil. ¹H NMR (500 MHz, CDCl3) δ: 0.15 (9 H, s), 1.40 (12 H, s), 4.07−4.09 (2 H, m), 4.17−4.20 (2 H, m), 4.57 (2 H, s), 5.98 (1 H, s), 8.10 (1 H, d, J = 1.5 Hz), 8.24 (1 H, d, $J = 1.5$ Hz). ¹³C NMR (125 MHz, CDCl₃) δ : −2.1, 24.9, 40.7, 65.3, 84.5, 103.5, 121.3, 132.9, 135.2, 136.9, 145.3. IR (CHCl₃): 1587 cm⁻¹. HRMS calcd for C₁₉H₃₀BN₃NaO₄Si (M + Na)⁺ m/z: 426.1996, found 426.1991.

5-(1,3-Dioxolan-2-yl)-1-methyl-1H-benzo[d]triazole (25). The mixture of *proximal*-10Cu (27 mg, 67 μ mol), 10% Pd/C (3.4 mg), and KF (12 mg, 0.20 mmol) was evacuated and backfilled with argon. MeOH (0.70 mL) was added into the reaction tube, and the mixture was heated using microwave for 3 h at 150 °C.²⁶ The reaction mixture was filtered through a short pad of Celite and evaporated under reduced pressure. CH_2Cl_2 was added into the [mi](#page-17-0)xture, filtered through a short pad of Celite, concentrated under reduced pressure to provide 25 as a colorless solid (13.5 mg), and used without further purification for the next reaction.

1-Methyl-1H-benzo[d]triazole-5-carbaldehyde (22). TsOH·- H₂O (3.8 mg, 20 μ mol) was added into an acetone solution (0.70 mL) of 25 (13.5 mg). The mixture was stirred at room temperature for 9 h. After the reaction, a saturated aqueous $NaHCO₃$ was added into the reaction mixture and extracted with EtOAc. The aqueous layer was extracted twice with EtOAc. The combined organic layers were washed with a saturated aqueous NaCl solution. The organic layer was dried over anhydrous $Na₂SO₄$ and evaporated under reduced pressure. The crude product was purified by flash column chromatography (hexane/EtOAc = 2:1) to provide the titled compound 22 (9.5 mg, 88% 2 steps) as a colorless solid. Mp 165−167 °C. ¹H NMR (500 MHz, CDCl₃) δ : 4.36 (3 H, s), 7.63 (1 H, d, J = 8.5 Hz), 8.09 (1 H, dd, J = 1.0, 8.5 Hz), 8.55 (1 H, brs), 10.13 (1 H, s). ¹³C NMR (125 MHz, CDCl₃) δ: 34.5, 110.1, 125.6, 126.1, 133.1, 136.5, 145.7, 191.1. IR (CHCl₃): 1701 cm⁻¹. HRMS calcd for C₈H₇N₃NaO (M + Na)⁺ m/ z: 184.0487, found 184.0481.

(4-Chlorophenyl)(1-methyl-1H-benzo[d]triazol-5-yl) methanol (26). An oven-dried flask was charged with 22 (46 mg, 0.29 mmol), capped with an inlet adapter with a three-way stopcock, then evacuated, and backfilled with argon. $Et₂O$ (2.8 mL, 0.10 M) was added via a syringe into the reaction mixture and cooled to −78 °C. 4- Chlorophenylmagnesiumbromide (1.0 M in Et₂O, 0.57 mL, 0.57 mmol) was added into the solution and stirred at −78 °C for 1 h. The reaction mixture was quenched with a saturated aqueous $NH₄Cl$ and extracted with EtOAc. The aqueous layer was extracted twice with EtOAc. The combined organic layers were washed with a saturated aqueous NaCl solution. The organic layer was dried over anhydrous Na2SO4 and evaporated under reduced pressure. The crude product was purified by column chromatography (hexane/EtOAc = $2:1$) to provide the titled compound 26 $(65 \text{ mg}, 83\%)$ as a colorless oil. ^1H NMR (500 MHz, CDCl3) δ: 4.28 (3 H, s), 6.00 (1 H, s), 7.32 (4 H, m), 7.46 (2 H, s), 8.08 (1 H, s). ¹³C NMR (125 MHz, CDCl₃) δ : 34.3, 75.3, 109.5, 117.3, 126.6, 127.9, 128.7, 133.1, 133.5, 139.9, 142.0, 145.9. IR (CHCl₃): 3603, 1600, 1491, 1281, 1236 cm⁻¹. HRMS calcd for $C_{14}H_{13}ClON_3$ $(M + H)^+$ m/z : 274.0747, found 274.0726.

5-((4-Chlorophenyl)(1H-1,2,4-triazol-1-yl)methyl)-1-methyl-1H-benzo[d]triazole (23). A mixture of 26 (65 mg, 0.23 mmol), 1,2,4-triazole (33 mg, 0.47 mmol), and TsOH·H₂O (13 mg, 71 μ mol) in a pear-shaped flask was dissolved in toluene (5.0 mL). The flask was connected with a Dean−Stark apparatus and stirred at 150 °C for 24 h. The reaction mixture was cooled to room temperature and quenched with a saturated aqueous $NaHCO₃$. The reaction mixture was

extracted with EtOAc, and the aqueous layer was extracted twice with EtOAc. The combined organic layers were washed with a saturated aqueous NaCl solution and dried over anhydrous $Na₂SO₄$. The solution was filtered through a glass filter and concentrated under reduced pressure. The crude product was purified by column chromatography (hexane/EtOAc = 1:2) to provide the titled compound 23 (60 mg, 79%) as a colorless oil. ¹H NMR (500 MHz, CDCl₃) δ : 4.31 (3 H, s), 6.90 (1 H, s), 7.10 (2 H, d, J = 8.0 Hz), 7.36 $(1 H, d, J = 8.0 Hz)$, 7.37 $(2 H, d, J = 8.0 Hz)$, 7.55 $(1 H, d, J = 8.0$ Hz), 7.78 (1H, s), 8.00 (1 H, s), 8.05 (1 H, s). 13C NMR (125 MHz, CDCl3) δ: 34.4, 66.9, 110.14, 119.8, 127.6, 129.3, 129.4, 133.4, 133.8, 135.0, 143.5, 145.9, 152.5. IR (CHCl₃): 1732, 1597, 1492, 1274, 1238 cm⁻¹. HRMS calcd for $C_{16}H_{14}CIN_6(M + H)^+ m/z$: 325.0968, found 325.0984.

General Procedure SIII: Synthesis of Benzyne Precursor Candidate (13B, 14A, 14B, and 33−37).

An oven-dried round-bottom flask was charged with 32^{14b} (crude product from 29) (1.0 equiv), capped with an inlet adapter with a three-way stopcock, then evacuated, and backfilled wit[h n](#page-17-0)itrogen. Anhydrous CH_2Cl_2 (0.30 M) was added via a syringe, and the mixture was cooled to 0 °C. Et₃N (3.0 equiv) and chlorinated benzene sulfonyl chloride (1.5 equiv) were added and stirred for several hours at room temperature. After the completion of the reaction, water was added to the reaction mixture. The mixture was extracted with CH_2Cl_2 , and the aqueous layer was extracted twice with $\mathrm{CH_2Cl_2}.$ The combined organic layers were washed with a saturated aqueous NaCl. The organic layer was dried over anhydrous $Na₂SO₄$, filtered through a glass filter, and concentrated under reduced pressure. The crude product was purified by flash column chromatography on silica gel to provide 13B, 14A, 14B, and 33−37.

2-Iodo-4-methyl-6-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)phenyl 4-Chlorobenzenesulfonate (13B). Following the general procedure SIII, a mixture of 2-iodo-4-methyl-6-(4,4,5,5 tetramethyl-1,3,2-dioxaborolan-2-yl)phenol 32B^{14b} (crude product from 29B) (1.0 g, 2.8 mmol), Et_3N (0.80 mL, 5.6 mmol), and 4chlorobenzenesulfonyl chloride (0.89 g, 4.2 m[mol\)](#page-17-0) in CH_2Cl_2 (5.0 mL, 0.30 M) was stirred for 5 h at room temperature. The crude product was purified by column chromatography (hexane/EtOAc = 11:1) to provide the titled compound 13B (0.85 g, 59% from 29B) as a colorless solid. Mp 165−170 °C. ¹H NMR (500 MHz, CDCl₃) δ : 1.41 (12 H, s), 2.29 (3 H, s), 7.48 (2 H, d, J = 8.5 Hz), 7.55 (1 H, s) 7.57 (1 H, s), 7.74 (2 H, d, \vec{J} = 8.5 Hz). ¹³C NMR (125 MHz, CDCl₃) δ: 20.2, 24.9, 84.6, 89.6, 129.5, 130.5, 135.3, 136.5, 138.3, 141.1, 142.7, 150.8. IR (CHCl₃): 3036 cm⁻¹. HRMS calcd for C₁₉H₂₁BClINaO₅S</sub> $(M + Na)^+$ m/z: 556.9834, found 556.9860.

2-Iodo-4-methyl-6-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)phenyl 2,5-Dichlorobenzenesulfonate (33). Following the general procedure SIII, a mixture of a mixture of $32B^{14b}$ (crude product from $29B$) (2.0 g, 5.6 mmol), Et₃N (1.6 mL, 11 mmol), and 2,5-dichlrobenzenesulfonyl chloride (1.6 mg, 6.7 mmol) [in](#page-17-0) CH_2Cl_2 (19 mL, 0.30 M) was stirred for 5 h at room temperature. The crude product was purified by column chromatography (hexane/EtOAc = $20:1$) to provide the titled compound 33 (2.5 g, 78% from 29B) as a colorless solid. Mp 115−117 °C. ¹H NMR (500 MHz, CDCl₃) δ: 1.40 $(12 \text{ H}, \text{s})$, 2.29 $(3 \text{ H}, \text{s})$, 7.46–7.54 $(2 \text{ H}, \text{m})$, 7.56 $(1 \text{ H}, \text{d}, J = 1.5 \text{ Hz})$, 7.61 (1 H, d, J = 1.5 Hz), 7.88 (1 H, d, J = 2.5 Hz). 13C NMR (125 MHz, CDCl₃) δ: 20.1, 24.8, 84.6, 88.7, 131.7, 133.0, 133.1, 133.4, 134.8, 136.7, 137.3, 138.4, 143.0, 151.5. IR (CHCl₃): 3036 cm⁻¹ . HRMS calcd for $C_{19}H_{20}BCl_2NaO_5S(M + Na)^+ m/z$: 590.9444, found 590.9447.

2-Iodo-4-methyl-6-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)phenyl 2,3-Dichlorobenzenesulfonate (34). Following the general procedure SIII, a mixture of a mixture of $32B^{14b}$ (crude product from 29B) (101 mg, 0.28 mmol), Et₃N (0.12 mL, 0.84 mmol), and 2,3-dichlrobenzenesulfonyl chloride (103 mg, [0.42](#page-17-0) mmol) in CH₂Cl₂ (1.0 mL, 0.30 M) was stirred for 3 h at room temperature. The crude product was purified by column chromatography (hexane/ EtOAc = $10:1$) to provide the titled compound 34 (126 mg, 79% from 29B) as a colorless solid. Mp 135−140 °C. ¹H NMR (500 MHz, CDCl₃) δ : 1.40 (12 H, s), 2.28 (3 H, s), 7.29 (1 H, t, J = 8.0 Hz), 7.56 $(1 H, s)$, 7.58 $(1 H, s)$, 7.74 $(1 H, d, J = 8.0 Hz)$, 7.78 $(1 H, d, J = 8.0 Hz)$ Hz). ¹³C NMR (125 MHz, CDCl₃) δ: 20.1, 24.8, 84.6, 88.5, 127.1, 130.2, 133.4, 135.6, 136.2, 136.6, 138.2, 138.3, 143.0, 151.7. IR (CHCl₃): 2982 cm⁻¹. HRMS calcd for C₁₉H₂₀BCl₂INaO₅S (M + Na)⁺ m/z: 590.9444, found 590.9451.

2-Iodo-4-methyl-6-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)phenyl 2,4-Dichlorobenzenesulfonate (35). Following the general procedure SIII, a mixture of a mixture of 32B^{14b} (crude product from $29B$) (101 mg, 0.30 mmol), Et₃N (0.12 mL, 0.84 mmol), and 2,4-dichlrobenzenesulfonyl chloride (103 mg, [0.42](#page-17-0) mmol) in CH_2Cl_2 (1.0 mL, 0.30 M) was stirred for 20 h at room temperature. The crude product was purified by column chromatography (hexane/ EtOAc = 10:1) to provide the titled compound 35 (116 mg, 73% from 29B) as a colorless solid. Mp 100−105 °C. ¹ H NMR (500 MHz, CDCl₃) δ : 1.40 (12 H, s), 2.28 (3 H, s), 7.33 (1 H, dd, J = 2.5, 8.5 Hz), 7.55 (1 H, d, J = 1.5 Hz), 7.57 (1 H, d, J = 2.5 Hz), 7.59 (1 H, d, J $= 1.5$ Hz), 7.78 (1 H, d, J = 8.5 Hz). ¹³C NMR (125 MHz, CDCl₃) δ : 20.1, 24.9, 84.6, 88.7, 127.3, 132.2, 132.9, 134.7, 135.9, 136.6, 138.4, 141.0, 142.9, 151.5. IR (CHCl₃): 2982 cm⁻¹. HRMS calcd for $C_{19}H_{20}BCl_2INaO_5S (M + Na)^+ m/z$: 590.9444, found 590.9471.

2-Iodo-4-methyl-6-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)phenyl 2,6-Dichlorobenzenesulfonate (36). Following the general procedure SIII, a mixture of a mixture of 32B^{14b} (crude product from 29B) (101 mg, 0.28 mmol), Et_3N (0.12 mL, 0.84 mmol), and 2,6-dichlrobenzenesulfonyl chloride (103 mg, [0.42](#page-17-0) mmol) in CH_2Cl_2 (1.0 mL, 0.30 M) was stirred for 5 h at room temperature. The crude product was purified by column chromatography (hexane/ EtOAc = 10:1) to provide the titled compound 36 (142 mg, 89% from 29B) as a colorless solid. Mp 197−202 °C. ¹ H NMR (500 MHz, CDCl₃) δ : 1.39 (12 H, s), 2.29 (3 H, s), 7.38 (1 H, t, J = 8.0 Hz), 7.46 $(2 H, d, J = 8.0 Hz)$, 7.56 $(1 H, d, J = 2.0 Hz)$, 7.61 $(1 H, d, J = 2.0$ Hz). ¹³C NMR (125 MHz, CDCl₃) δ: 20.1, 24.8, 84.6, 88.2, 131.5, 133.6, 135.1, 136.7, 137.2, 138.3, 142.9, 151.9. IR (CHCl₃): 3011 cm⁻¹. HRMS calcd for C₁₉H₂₀BCl₂INaO₅S (M + Na)⁺ m/z: 590.9444, found 590.9468.

2-Iodo-6-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl) phenyl 2,4,5-Trichlorobenzenesulfonate (14A). Following the general procedure SIII, a mixture of a mixture of 2-iodo-6-(4,4,5,5 tetramethyl-1,3,2-dioxaborolan-2-yl)phenol $32A^{14b}$ (crude product from 29A) (2.9 mmol of 29A), Et_3N (0.87 mL, 6.2 mmol), and 2,4,5-trichlrobenzenesulfonyl chloride (0.87 g, 3[.1 m](#page-17-0)mol) in CH_2Cl_2 (7.0 mL) was stirred for 4 h at room temperature. The crude product was purified by column chromatography (hexane/EtOAc = $10:1$) to provide the titled compound 14A (1.0 g, 75% from 29A) as a colorless σ solid. Mp 106−108 °C. ¹H NMR (500 MHz, CDCl₃) δ : 1.39 (12 H, s), 7.02 (1 H, t, $J = 7.5$ Hz), 7.67 (1 H, s), 7.79 (1 H, dd, $J = 1.5$, 7.5 Hz), 7.81 (1 H, dd, J = 1.5, 7.5 Hz), 7.97 (1 H, s). 13C NMR (125 MHz, CDCl₃) δ: 20.1, 24.8, 84.7, 88.7, 131.6, 133.0, 133.4, 133.5, 135.6, 136.8, 138.6, 139.3, 143.0, 151.4. IR (CHCl₃): 3009 cm[−] . HRMS calcd for $C_{18}H_{17}BCl_3NaO_5S (M + Na)^+ m/z$: 610.8898, found 610.8871.

2-Iodo-4-methyl-6-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)phenyl 2,4,5-Trichlorobenzenesulfonate (14B). Following the general procedure SIII, a mixture of a mixture of $32B^{14b}$ (crude product from 29B) (1.1 g, 3.0 mmol), Et₃N (1.2 mL, 8.9 mmol), and 2,4,5-trichlrobenzenesulfonyl chloride (1.2 g, 4.4 mmol) in CH_2Cl_2 (10 mL, 0.30 M) was stirred for 12 h at room temperature. The crude product was purified by column chromatography (hexane/EtOAc = 10:1) to provide the titled compound $14B$ (1.3 g, 75% from 29B) as a colorless solid. Mp 134−138 °C. ¹H NMR (500 MHz, CDCl₃) δ: 1.39

 $(12 \text{ H}, \text{s})$, 2.30 $(3 \text{ H}, \text{s})$, 7.57 $(1 \text{ H}, \text{d}, I = 1.5 \text{ Hz})$, 7.62 $(1 \text{ H}, \text{d}, I = 1.5 \text{ Hz})$ Hz), 7.67 (1 H, s), 7.97 (1 H, s). ¹³C NMR (125 MHz, CDCl₃) δ : 20.1, 24.8, 84.7, 88.7, 131.6, 133.0, 133.4, 133.5, 135.6, 136.8, 138.6, 139.3, 143.0, 151.4. IR (CHCl₃): 3034 cm⁻¹. HRMS calcd for $C_{19}H_{19}BCl_3NaO_5S (M + Na)^+ m/z$: 624.9054, found 624.9068.

Methyl 3-Iodo-5-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2 yl)-4-(((2,4,5-trichlorophenyl)sulfonyl)oxy)benzoate (37). Following the general procedure SIII, a mixture of methyl 4-hydroxy-3 iodo-5-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)benzoate 32D14b (1.06 g, crude product from 29D), $Et₃N$ (1.1 mL, 7.9 mmol), and 2,4,5-trichlrobenzenesulfonyl chloride (1.1 g, 3.9 mmol) in CH_2Cl_2 (26 mL, 0.10 M) was stirred for 12 h at room temperature. The crude product was purified by column chromatography (hexane/EtOAc $=$ 5:1) to provide the titled compound 37 (1.46 g, 53% from 29D) as a colorless solid. Mp 174−176 °C. ¹H NMR (500 MHz, CDCl₃) δ: 1.40 (12 H, s), 3.92 (3 H, s), 7.68 (1 H, brs), 7.98 (1 H, brs), 8.42 (1 H, d, $J = 2.0$ Hz), 8.47 (1 H, d, $J = 2.0$ Hz). ¹³C NMR (125 MHz, CDCl₃) δ : 24.8, 52.5, 85.0, 88.9, 129.9, 131.8, 132.9, 133.3, 133.6, 135.4, 137.4, 139.7, 143.8, 156.5, 164.6. IR (CHCl₃): 1705 cm⁻¹. HRMS calcd for $C_{20}H_{19}BCl_3NaO_7S (M + Na)^+ m/z$: 668.8952, found 668.8944.

4-Formyl-2-iodo-6-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)phenyl 2,4,5-Trichlorobenzenesulfonate (38). An ovendried round-bottom flask was charged with 37 (1.34 g, 2.1 mmol), capped with an inlet adapter with a three-way stopcock, then evacuated, and backfilled with nitrogen. CH_2Cl_2 (11 mL) was added into the flask via a syringe and cooled to −78 °C. A 1.0 M diisobutyl aluminum hydride solution in CH₂Cl₂ (2.1 mL, 2.1 mmol) was added to the mixture and stirred over 1 h. MeOH (0.50 mL) and a saturated aqueous Rochell's salt (2.0 mL) were added to the reaction mixture and stirred over 1 h at room temperature. The mixture was filtered through a pad of Celite and constructed twice by CH_2Cl_2 . The combined organic layer was dried over anhydrous $Na₂SO₄$, and solvent was removed under reduced pressure. Compound 38 was obtained as a colorless solid (1.26 g) and used as such without further purification for the next reaction.

4-(1,3-Dioxolan-2-yl)-2-iodo-6-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)phenyl 2,4,5-Trichlorobenzenesulfonate (14C). A mixture of 38 (196 mg) and ethylene glycol (40 μ L, 0.72 mmol) in CH_2Cl_2 (3.0 mL) was cooled to 0 °C and stirred under argon. Trimethylsilyl chloride (80 μ L, 0.63 mmol) was added to the solution and slowly warmed to room temperature. After 9 h, solvent was removed under reduced pressure. The crude product was purified by column chromatography (hexane/EtOAc = 6:1) to provide the titled compound 14C (104 mg, 49%) as a colorless solid. Mp 73−75 °C. ¹H NMR (500 MHz, CDCl₃) δ: 1.39 (12 H, s), 4.03–4.11 (4 H, m), 5.76 (1 H, s), 7.67 (1 H, brs), 7.86 (1 H, d, J = 2.0 Hz), 7.92 (1 H, d, J = 2.0 Hz), 7.96 (1 H, brs). ¹³C NMR (125 MHz, CDCl₃) δ : 24.8, 65.4, 84.8, 88.8, 101.8, 131.7, 133.0, 133.4, 133.5, 134.5, 135.6, 138.3, 139.4, 140.6, 153.9. IR (CHCl₃): 1394 cm⁻¹. HRMS calcd for $C_{21}H_{21}BCl_3INaO_7S (M + Na)^+ m/z$: 682.9109, found 682.9110.

Synthesis of Silylbenzyne Precursors (16B' and 16E')^{14c} 2-(tert-Butyldimethylsilyl)-4-methyl-6-(trimethylsilyl)phenol (39). An oven-dried round-bottom flask was charged with 2-br[omo](#page-17-0)-6- (tert-butyldimethylsilyl)-4-methylphenol^{14a} (3.6 g, 12 mmol), capped with an inlet adapter with a three-way stopcock, then evacuated, and backfilled with nitrogen. CH₂Cl₂ (30 m[L, 0](#page-17-0).40 M) was added into the flask via a syringe, cooled to 0 °C, and stirred for 10 min. Triethylamine (2.0 mL, 14 mmol) and trimethylsilyl chloride (1.9 mL, 14 mmol) were added into the flask, and the mixture was stirred at room temperature for 3 h. The reaction mixture was evaporated under reduced pressure. Hexane was added to the residue, filtered through a pad of short pad of Celite, and concentrated under reduced pressure to afford (3-bromo-5-methyl-2-[(trimethylsilyl)oxy]phenyl)- (tert-butyl)dimethylsilane. THF (24 mL, 0.50 M) was added to the crude product, and the mixture was stirred at −78 °C for 10 min. n-BuLi (1.6 M in hexane, 7.4 mL, 12 mmol) was added to the mixture and stirred at room temperature for 1 h. The reaction mixture was quenched with a saturated aqueous NH4Cl and evaporated under reduced pressure. The residue was extracted with hexane and water. The aqueous layer was extracted twice with hexane. The combined organic layers were washed with a saturated aqueous NaCl solution. The organic layer was dried over anhydrous $MgSO₄$, and solvent was removed under reduced pressure. The crude product was purified by column chromatography (hexane) to provide the titled compound 39 (3.2 g, 92%) as a colorless solid. Mp 54−56 °C. ¹ H NMR (500 MHz, CDCl₃) δ : 0.31 (9 H, s), 0.36 (6 H, s), 0.91 (9 H, s), 2.27 (3 H, s), 4.90 (1 H, s), 7.12 (1 H, d, J = 2.0 Hz), 7.17 (1 H, d, J = 2.0 Hz). ¹³C NMR (125 MHz, CDCl₃) δ: −4.6, −0.8, 17.6, 20.6, 26.6, 120.5, 124.6, 128.5, 137.2, 138.2, 163.5 . IR (CHCl₃): 3612, 1570, 1464, 1402, 1253, 1167 cm⁻¹. HRMS calcd for C₁₆H₂₉OSi₂ (M-H⁺) m/z: 293.1733, found 293.1757.

2-(tert-Butyldimethylsilyl)-4-(1,3-dioxolan-2-yl)-6- (trimethylsilyl)phenyl Trifluoromethanesulfonate (16B′). An oven-dried round-bottom flask was charged with 39 (1.0 g, 3.4 mmol), capped with an inlet adapter with a three-way stopcock, then evacuated, and backfilled with nitrogen. Et_2O (14 mL, 0.25 M) was added into the flask via a syringe, cooled to -78 °C, and stirred for 10 min. n-BuLi (1.62 M in hexane, 2.3 mL, 3.7 mmol) was added and stirred at −78 °C for 10 min. Then, trifluoromethanesulfonic anhydride (0.62 mL, 3.7 mmol) was added into the mixture at -78 °C. The mixture was warmed to room temperature and stirred at room temperature for 3 h. The reaction mixture was quenched with a saturated aqueous NH₄Cl and extracted with EtOAc. The aqueous layer was extracted twice by EtOAc. The combined organic layers were washed with a saturated aqueous NaCl solution. The organic layer was dried over anhydrous $Na₂SO₄$ and concentrated under reduced pressure (this reaction did not complete). Et₂O (13 mL, 0.25 M) was added into the flask via a syringe, cooled to −78 °C, and stirred for 10 min. n-BuLi (1.62 M in hexane, 0.23 mmol) was added and stirred at −78 °C for 10 min. Then, trifluoromethanesulfonic anhydride (63 μ L, 0.38 mmol) was added into the mixture at −78 °C. The mixture was warmed to room temperature and stirred at room temperature for 3 h. The reaction mixture was quenched with a saturated aqueous $NH₄Cl$ and extracted with EtOAc. The aqueous layer was extracted twice by EtOAc. The combined organic layers were washed with a saturated aqueous NaCl solution. The organic layer was dried over anhydrous Na2SO4 and concentrated under reduced pressure. The crude product was purified by column chromatography (hexane/EtOAc = $20:1$) to provide the titled compound $16B^\prime \, (1.31 \text{ g}, 90 \text{\%})$ as a colorless oil. ^1H NMR (500 MHz, CD₃OD) δ: 0.33 (9 H, s), 0.39 (6 H, s), 0.78 (9 H, s), 2.39 (3 H, s), 7.43 (1 H, d, J = 2.0 Hz), 7.47 (1 H, d, J = 2.0 Hz). ¹³C NMR (125 MHz, CDCl₃) δ: −4.0, −0.56, 18.1, 20.8, 26.9, 118.5 $(q, J = 318 \text{ Hz})$, 131.6, 134.8, 136.1, 138.7, 139.2, 152.7. IR $(CHCl₃)$: 1566, 1463, 1392, 1367, 1253 cm⁻¹. HRMS calcd for C₁₇H₃₀F₃O₃SSi₂ $(M + H)^+$ m/z: 427.1406, found 427.1419.

4-(1,3-Dioxolan-2-yl)-2,6-diiodophenol (40). A mixture of 4 hydroxy-3,5-diiodobenzaldehyde (3.0 g, 8.0 mmol), ethylene glycol $(0.90 \text{ mL}, 16 \text{ mmol})$ and TsOH·H₂O $(69 \text{ mg}, 0.40 \text{ mmol})$ was dissolved in toluene. The reaction flask was connected with Dean− Stark apparatus and stirred at 150 °C for 10 h. After the reaction was completed, K_2CO_3 (0.20 g, 1.4 mmol) was added to the mixture, filtered, and concentrated under reduced pressure. The crude product was purified by column chromatography (hexane/EtOAc = $1:1$) to provide the titled compound 40 (3.0 g, 89%) as a colorless solid. Mp 120−122 °C. ¹ H NMR (500 MHz, CDCl3) δ: 4.04 (4 H, m), 5.66 (1 H, s), 5.88 (1 H, s), 7.77 (2 H, s). ¹³C NMR (125 MHz, CDCl₃) δ : 65.2, 82.0, 101.4, 133.9, 137.5, 154.1. IR (CHCl₃): 3479, 2891, 1600, 1460, 1363, 1238 cm⁻¹. HRMS calcd for $C_9H_9I_2O_3$ $(M + H)^+$ m/z: 418.8670, found 418.8641.

(4-(1,3-Dioxolan-2-yl)-2,6-diiodophenoxy)(tert-butyl) dimethylsilane (41). A mixture of 40 (1.0 g, 2.4 mmol) and imidazole (0.24 g, 3.6 mmol) was dissolved in THF (10 mL, 0.25 M). tert-Butyldimethylsilyl chloride was added to the mixture and stirred at room temperature for 1 h. The reaction mixture was quenched with a saturated aqueous NH₄Cl and extracted with EtOAc. The aqueous layer was extracted twice with EtOAc. The combined organic layers were washed with a saturated aqueous NaCl solution. The organic layer was dried over anhydrous $Na₂SO₄$ and concentrated under reduced pressure. The crude product was purified by column chromatography (hexane/EtOAc = 10:1) to provide the titled

compound 41 (1.20 g, 95%) as a colorless solid. Mp 103−104 °C. ¹H NMR (500 MHz, CDCl₃) δ: 0.46 (6 H, s), 1.07 (9 H, s), 4.05 (4 H, m), 5.65 (1 H, s), 7.86 (2 H, s). ¹³C NMR (125 MHz, CDCl₃) δ: 0.0, 18.9, 26.5, 65.3, 88.5, 101.5, 134.4, 138.5, 155.9. IR (CHCl₃): 2958, 2931, 2887, 2858, 1600, 1448, 1276, 1255 cm⁻¹. HRMS calcd for $C_{15}H_{23}I_{2}O_{3}Si$ $(M + H)^{+}m/z$: 532.9528, found 532.9506.

2-(tert-Butyldimethylsilyl)-4-(1,3-dioxolan-2-yl)-6-iodophenol (42). An oven-dried round-bottom flask was charged with 41 (0.80 g, 1.51 mmol), capped with an inlet adapter with a three-way stopcock, then evacuated, and backfilled with nitrogen. THF (15 mL, 0.10 M) was added into the flask via a syringe, cooled to −78 °C, and stirred for 10 min. n-BuLi (1.65 M in hexane, 1.2 mL, 2.0 mmol) was added to the solution at -78 °C, warmed to room temperature, and stirred for 2 h. The reaction mixture was quenched with a saturated aqueous NH4Cl and extracted with EtOAc. The aqueous layer was extracted twice with EtOAc. The combined organic layers were washed with a saturated aqueous NaCl solution. The organic layer was dried over anhydrous Na₂SO₄ and concentrated under reduced pressure. The crude product was purified by column chromatography (hexane/EtOAc = $10:1$) to provide the titled compound 42 as a colorless solid (0.53 g, 88%). Mp 112−113 °C. ¹ H NMR (500 MHz, CDCl3) δ: 0.30 (6 H, s), 0.89 (9 H, s), 0.78 (9 H, s), 4.06 (4 H, m), 5.53 (1 H, s), 5.70 (1 H, s), 7.36 (1 H, d, $J = 2.0$ Hz), 7.80 (1 H, d, $J =$ 2.0 Hz). ¹³C NMR (125 MHz, CDCl₃) δ : −4.8, 17.6, 27.0, 65.2, 87.2, 102.8, 123.5, 131.2, 135.7, 137.4, 159.4. IR (CHCl₃): 3489, 2955, 2928, 2886, 2856, 1589, 1560, 1413, 1359, 1248 cm[−]¹ . HRMS calcd for $C_{15}H_{24}IO_3Si$ $(M + H^+)$ m/z : 407.0539, found 407.0551.

2-(tert-Butyldimethylsilyl)-4-(1,3-dioxolan-2-yl)-6- (trimethylsilyl)phenol (43). An oven-dried round-bottom flask was charged with 42 (63 mg, 0.15 mmol), capped with an inlet adapter with a three-way stopcock, then evacuated, and backfilled with nitrogen. CH₂Cl₂ (1.0 mL, 0.25 M) was added into the flask via a syringe, cooled to 0 °C, and stirred for 10 min. Triethylamine (26 μ L, 0.19 mmol) and trimethylsilyl chloride (25 μ L, 0.19 mmol) were added into the flask, and the mixture was stirred at room temperature for 1 h. The reaction mixture was evaporated under reduced pressure. Hexane was added to the residue, filtered through a pad of short pad of Celite and concentrated under reduced pressure to afford (5-(1,3 dioxolan-2-yl)-3-iodo-2-((trimethylsilyl)oxy)phenyl)(tert-butyl) dimethylsilane. The crude product was dissolved in THF (1.0 mL, 0.25 M) and stirred at −78 °C for 10 min. n-BuLi (1.5 M in hexane, 0.12 mL, 0.19 mmol) was added to the mixture and stirred at room temperature for 1 h. The reaction mixture was quenched with a saturated aqueous NH4Cl and evaporated under reduced pressure. The residue was extracted with hexane and water. The aqueous layer was extracted twice with hexane. The combined organic layers were washed with a saturated aqueous NaCl solution. The organic layer was dried over anhydrous $Na₂SO₄$ and solvent was removed under reduced pressure. The crude product was purified by column chromatography (hexane/EtOAc = 20:1) to provide the titled compound 43 (23 mg, 42%) as a colorless oil. ¹H NMR (500 MHz, CDCl₃) δ : 0.32 (9 H, s), 0.39 (6 H, s), 0.92 (9 H, s), 4.08 (4 H, m), 5.16 (1 H, s), 5.74 (1 H, s), 7.41 (1 H, d, $J = 2.0$ Hz), 7.50 (1 H, d, $J = 2.0$ Hz). ¹³C NMR (125 MHz, CDCl₃) δ: −4.6, −0.9, 17.6, 26.5, 65.2, 104.2, 120.3, 125.0, 128.6, 134.9, 136.3, 166.5. IR (CHCl₃): 3609, 2955, 1587, 1576, 1409, 1364, 1238 cm⁻¹. HRMS calcd for C₁₈H₃₃O₃Si₂ (M + H⁺) m/z: 353.1968, found 353.1982.

2-(tert-Butyldimethylsilyl)-4-methyl-6-(trimethylsilyl)phenyl 1,1,2,2,3,3,4,4,4-Nonafluorobutane-1-sulfonate (16E′). Compound 43 (4.4 g, 12.5 mmol) and 18-crown-6 (3.3 g, 12.5 mmol) were dissolved in THF (125 mL, 0.10 M) and stirred at 0 °C for 10 min. NaH (0.75 g, 18.7 mmol) was added to the mixture and stirred at 0 \degree C for 30 min. Then, nonafluorobutanesulfonyl fluoride²³ (0.62 mL, 3.7 mmol) was added to the solution and stirred at 80 °C for 20 h. The reaction mixture was quenched with water and evap[ora](#page-17-0)ted under reduced pressure. The residue was extracted with EtOAc and water, and the aqueous layer was extracted twice with hexane. The combined organic layers were washed with a saturated aqueous NaCl solution. The organic layer was dried over anhydrous $Na₂SO₄$, filtered with a glass filter, and concentrated under reduced pressure (this reaction did not complete). THF (125 mL, 0.10 M), NaH (0.25 g, 6.2 mmol) and nonafluorobutanesulfonyl fluoride (1.2 mL, 7.1 mmol) were added again to obtain a full conversion. The mixture was stirred at 80 °C for 20 h. The reaction mixture was quenched with water and evaporated under reduced pressure. The residue was extracted with EtOAc and water, and the aqueous layer was extracted twice with hexane. The combined organic layers were washed with a saturated aqueous NaCl solution. The organic layer was dried over anhydrous $Na₂SO₄$, filtered with a glass filter and evaporated under reduced pressure. The crude product was purified by column chromatography (hexane/ $EtOAc =$ 5:1) to provide the titled compound 16E′ (6.1 g, 77%) as a colorless solid. Mp 67–69 °C. ¹H NMR (500 MHz, CDCl₃) δ: 0.35 (9 H, s), 0.40 (6 H, s), 0.79 (9 H, s), 4.05−4.12 (4 H, m), 5.84 (1 H, s), 7.67 (2 H, s). ¹³C NMR (125 MHz, CDCl₃) δ : −4.0, 0.5, 18.1, 26.9, 65.3, 102.9, 108.6−120.8 (4 C, m), 132.4, 135.5, 135.9, 136.1, 136.3, 136.7, 155.2. IR (CHCl₃): 1603, 1396, 1352, 1238, 1194, 1146 cm⁻¹. HRMS calcd for $C_{25}H_{28}F_9O_3SSi_2$ $(M + H)^+$ m/z : 635.1154, found 635.1149.

■ ASSOCIATED CONTENT

S Supporting Information

Optimization of 3-borylbenzyne precursors (Tables S1−S3), interaction of ions with silyl- and borylbenzynes 4 and 5 (Tables S4–S6), ¹H and ¹³C NMR spectra for all new compounds, and Cartesian Coordinates for ground and transition state structures. This material is available free of charge via the Internet at http://pubs.acs.org.

■ AUTHOR INFORM[ATION](http://pubs.acs.org)

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Notes

The auth[ors declare no competing](mailto:akai@u-shizuoka-ken.ac.jp) fi[nancial interest.](mailto:tokiwa@rikkyo.ac.jp)

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■ REFERENCES

(1) (a) Eicher, T.; Hauptmann, S. The Chemistry of Heterocycles, Wiley-VCH: Weinheim, 2003. (b) Katrizky, A. R.; Pozharskii, A. F. Handbook of Heterocyclic Chemistry, 2nd ed., Pergamon: New York, 2000.

(2) Saberi, M. R.; Vinh, T. K.; Yee, S. W.; Griffiths, B. J. N.; Evans, P. J.; Simons, C. J. Med. Chem. 2006, 49, 1016−1022.

(3) Hayashi, S.; Hirao, A.; Imai, A.; Nakamura, H.; Murata, Y.; Ohashi, K.; Nakata, E. J. Med. Chem. 2009, 52, 610−625.

(4) Patel, D.; Jain, M.; Shah, S. R.; Bahekar, R.; Jadav, P.; Darji, B.; Siriki, Y.; Bandyopadhyay, D.; Joharapurkar, A.; Kshirsagar, S.; Patel, H.; Shaikh, M.; Sairam, K. V. V. M.; Patel, P. ChemMedChem 2011, 6, 1011−1016.

(5) Dixit, P. P.; Nair, P. S.; Patil, V. J.; Jain, S.; Arora, S. K.; Sinha, N. Bioorg. Med. Chem. Lett. 2005, 15, 3002−3005.

(6) Rezaei, Z.; Khabnadideh, S.; Pakshir, K.; Hossaini, Z.; Amiri, F.; Assadpour, E. Eur. J. Med. Chem. 2009, 44, 3064−3067.

(7) Dubey, A.; Srivastava, S. K.; Srivastava, S. D. Bioorg. Med. Chem. Lett. 2011, 21, 569−573.

(8) Semple, G.; Skinner, P. J.; Cherrier, M. C.; Webb, P. J.; Sage, C. R.; Tamura, S. Y.; Chen, R.; Richman, J. G; Connolly, D. T. J. Med. Chem. 2006, 49, 1227−1230.

(9) Prous, J.; Graul, A.; Castañ er, J. Drugs Future 1994, 19, 457−459.

(10) (a) Matsumoto, T.; Sohma, T.; Hatazaki, S.; Suzuki, K. Synlett 1993, 843−846. (b) Jin, T.; Yamamoto, Y. Angew. Chem., Int. Ed. 2007, 46, 3323−3325. (c) Shi, F.; Waldo, J. P.; Chen, Y.; Larock, R. C. Org. Lett. 2008, 10, 2409−2412. (d) Liu, Z.; Shi, F.; Martinez, P. D. G.; Raminelli, C.; Larock, R. C. J. Org. Chem. 2008, 73, 219−226. (e) Dai, M.; Wang, Z.; Danishefsky, S. J. Tetrahedron Lett. 2008, 49, 6613−6616. (f) Campbell-Verduyn, L.; Elsinga, P. H.; Mirfeizi, L.; Dierckx, R. A.; Feringa, B. L. Org. Biomol. Chem. 2008, 6, 3461−3463. (g) Zhang, F.; Moses, J. E. Org. Lett. 2009, 11, 1587−1590. (h) Spiteri, C.; Keeling, S.; Moses, J. E. Org. Lett. 2010, 12, 3368−3371. (i) Dubrovskiy, A. V.; Larock, R. C. Org. Lett. 2010, 12, 1180−1183. (j) Li, P.; Zhao, J.; Wu, C.; Larock, R. C.; Shi, F. Org. Lett. 2011, 13, 3340−3343. (k) Fang, Y.; Wu, C.; Larock, R. C.; Shi, F. J. Org. Chem. 2011, 76, 8840−8851. (l) Lu, C.; Dubrovskiy, A. V.; Larock, R. C. J. Org. Chem. 2012, 77, 2279−2284.

(11) For recent reviews, see: (a) Pellissier, H.; Santelli, M. Tetrahedron 2003, 59, 701−730. (b) Wenk, H. H.; Winkler, M.; Sander, W. Angew. Chem., Int. Ed. 2003, 42, 502−528. (c) Dyke, A. M.; Hester, A. J.; Lloyd-Jones, G. C. Synthesis 2006, 24, 4093−4112. (d) Bhunia, A.; Yetra, S. R.; Biju, A. T. Chem. Soc. Rev. 2012, 41, 3140−3152. (e) Tadross, P. M.; Stoltz, B. M. Chem. Rev. 2012, 112, 3550−3577.

(12) Garg and Houk et al. and our group have independently studied the $(3 + 2)$ cycloaddition reactions of 3-silylbenzynes with 1,3-dipoles along with computational analyses, and both of our groups have come to essentially the same conclusions. See: Bronner, S. M.; Mackey, J. L.; Houk, K. N.; Garg, N. K. J. Am. Chem. Soc. 2012, 134, 13966−13969.

(13) Very recently, Garg et al. achieved regioselective $(3 + 2)$ cycloadditions of pyridynes possessing either a sulfamoyloxy group or a bromo group as a directing group; see: Goetz, A. E.; Garg, N. K. Nat. Chem. 2013, 5, 54−60.

(14) (a) Akai, S.; Ikawa, T.; Takayanagi, S.; Morikawa, Y.; Mohri, S.; Tsubakiyama, M.; Egi, M.; Wada, Y.; Kita, Y. Angew. Chem., Int. Ed. 2008, 47, 7673−7676. (b) Ikawa, T.; Takagi, A.; Kurita, Y.; Saito, K.; Azechi, K.; Egi, M.; Kakiguchi, K.; Kita, Y.; Akai, S. Angew. Chem., Int. Ed. 2010, 49, 5563−5566. (c) Ikawa, T.; Nishiyama, T.; Shigeta, T.; Mohri, S.; Morita, S.; Takayanagi, S.; Terauchi, Y.; Morikawa, Y.; Takagi, A.; Ishikawa, Y.; Fujii, S.; Kita, Y.; Akai, S. Angew. Chem., Int. Ed. 2011, 50, 5674−5677.

(15) A part of this work was reported in our very recent brief account; see: Ikawa, T.; Tokiwa, H.; Akai, S. J. Synth. Org. Chem. Jpn. 2012, 70, 1123−1133.

(16) Pansegrau, P. D.; Rieker, W. F.; Meyers, A. I. J. Am. Chem. Soc. 1988, 110, 7178−7184.

(17) (a) Sapountzis, I.; Lin, W.; Fischer, M.; Knochel, P. Angew. Chem., Int. Ed. 2004, 43, 4364−4366. (b) Lin, W.; Chen, L.; Knochel, P. Tetrahedron 2007, 63, 2787−2797.

(18) 2,4,5-Trichlorobenzenesulfonyl chloride, used for preparing 14B, is readily available from several chemical companies, e.g., TCI Fine Chemicals, TCI America, Acros Organics, Sigma-Aldrich, and Alfa Aesar.

(19) According to our previous work, $14b$ new borylbenzyne precuosors 14 were prepared from 2,6-diiodophenols 28. For details, see experimental section and reference 14b.

(20) According to our previous work,^{14c} silylbenzyne precursors 16 were prepared from 2,6-dibromophenols 44 as shown in the scheme bel[ow.](#page-6-0) [For](#page-6-0) [details,](#page-6-0) [see](#page-6-0) [re](#page-6-0)ference 14c.

(21) Allred, A. L.; Rochow, E. G. J. Inorg. Nucl. Chem. 1958, 5, 264− 268.

(22) Chattopadhyay, S.; Chattopadhyay, U.; Mathur, P. P.; Saini, K. S.; Ghosal, S. Planta Med. 1983, 49, 252−254.

(23) Ikawa, T.; Nishiyama, T.; Nosaki, T.; Takagi, A.; Akai, S. Org. Lett. 2011, 13, 1730−1733.

(24) Dhanak, D.; Knight, S. D. Patent WO 2007/103755 A2.

(25) Kim, S. W.; Biegon, A.; Katsamanis, Z. E.; Ehrlich, C. W.; Hooker, J. M.; Shea, C.; Muench, L.; Xu, Y.; King, P.; Carter, P.;

Alexoff, D. L.; Fowler, J. S. Nucl. Med. Biol. 2009, 36, 323−334.

(26) These deborylation conditions (soon to be published) were recently found in our laboratory.

(27) All DFT calculations were performed at the B3LYP/6-31G(d) level except for iodine atom (B3LYP/LanL2DZ) of 6k, TS17, and TS18. Optimized structures of the reactants, transition sates, and products were characterised by analytical frequency calculations, and all the total electronic energies were included in the zero-point energy corrections at the same level. The calculated number of imaginary frequencies (NImag) will determine whether the optimized structures are the energy minima (NImag = 0) or transition states (NImag = 1) along the reaction pathway. All calculations were carried out by using the Gaussian 09 revision A.02 (Frisch, M. J. et al. ; see the Supporting Information for full citation).

(28) The interaction of ions in the reaction mixture with benzynes (4 and 5) may have some influence on the regioselectivities. [However,](#page-16-0) [in](#page-16-0) [fact,](#page-16-0) [the](#page-16-0) [rat](#page-16-0)ios of distal- to proximal-adducts were not significantly affected by changing the conditions of preparation of 4 and 5 (see Tables S5 and S6 in Supporting Information). Therefore, we have reached the conclusion that the effects of ions were negligibly small, and they were not c[onsidered for the calcul](#page-16-0)ations of ground and transition state structures.

(29) (a) Cheong, P. H.-Y.; Paton, R. S.; Bronner, S. M.; Im, G.-Y. J.; Garg, N. K.; Houk, K. N. J. Am. Chem. Soc. 2010, 132, 1267−1269. (b) Im, G-Y.; Bronner, S. M.; Goetz, A. E.; Paton, R. S.; Cheong, P. H.-Y.; Houk, K. N.; Garg, N. K. J. Am. Chem. Soc. 2010, 132, 17933− 17944.

(30) In general, accurate entropy and thermodynamic corrections on theoretical reaction analysis should be required (see: Kruse, H.; Goerigk, L.; Grimme, S. J. Org. Chem. 2012, 77, 10824−10834. Grimme, S. ChemPhysChem 2012, 13, 1407−1409). However, ΔΔH[⧧] values of the $(3 + 2)$ cycloaddition reactions of borylbenzynes 4 without entropy contributions were applied for calculating the theoretical ratios of distal- to proximal-adducts because it is difficult to correctly evaluate the activation entropy for the reactions of 4 due to local interactions between the boryl group and solvent molecules. On the other hand, $\Delta \Delta G^{\ddagger}$ values of the $(3 + 2)$ cycloaddition reactions of silylbenzynes 5 were applied for the calculation.

(31) As for the $(3 + 2)$ cycloaddition of 3-silylbenzynes, similar discussion was also reported by Houk and Garg.¹

(32) All graphics were prepared with CYLview: CYLview, 1.0b; Legault, C. Y. Université de Sherbrooke, 2009 (http://www.cylview. org).

(33) (a) Miyaura, N.; Suzuki, A. Chem. Rev. 1995, 95, 2457−2483. (b) Monnier, F.; Taillefer, M. Angew. Chem., Int. Ed. 2009, 48[,](http://www.cylview.org) [6954](http://www.cylview.org)− [697](http://www.cylview.org)1.

(34) Maddani, M. R.; Moorthy, S. K.; Prabhu, K. R. Tetrahedron 2010, 66, 329−333.

(35) Hubbard, A.; Okazaki, T.; Laali, K. K. J. Org. Chem. 2007, 73, 316−319.

(36) Liu, C.-Y.; Knochel, P. J. Org. Chem. 2007, 72, 7106−7115.

(37) Nguyen, T.-T.-T.; Simon, F.-X.; Schmutz, M.; Mesini, P. J. ́ Chem. Commun. 2009, 3457−3459.

(38) Murali, A.; Puppala, M.; Varghese, B.; Baskaran, S. Eur. J. Org. Chem. 2011, 5297−5302.

(39) Lamani, M.; Prabhu, K. R. Angew. Chem., Int. Ed. 2010, 49, 6622−6625.

(40) Young, M. B.; Barrow, J. C.; Glass, K. L.; Lundell, G. F.; Newton, C. L.; Pellicore, J. M.; Rittle, K. E.; Selnick, H. G.; Stauffer, K. J.; Vacca, J. P.; Williams, P. D.; Bohn, D.; Clayton, F. C.; Cook, J. J.; Krueger, J. A.; Kuo, L. C.; Lewis, S. D.; Lucas, B. J.; McMasters, D. R.; Miller-Stein, C.; Pietrak, B. L.; Wallace, A. A.; White, R. B.; Wong, B.;

Yan, Y.; Nantermet, P. G. J. Med. Chem. 2004, 47, 2995−3008.

(41) Javed, M. I.; Brewer, M. Org. Synth. 2008, 85, 189−195.

(42) Katritzky, A. R.; Button, M. A. C.; Denisenko, S. N. J. Heterocycl. Chem. 2000, 37, 1505−1510.

(43) Cossy, J.; Tresnard, L.; Pardo, D. G. Eur. J. Org. Chem. 1999, 1925−1933.

(44) Still, W. C.; Kahn, M.; Mitra, A. J. Org. Chem. 1978, 43, 2923− 2925.

(45) Clary, J. W.; Rettenmaier, T. J.; Snelling, R.; Bryks, W.; Banwell, J.; Wipke, W. T.; Singaram, B. J. Org. Chem. 2011, 76, 9602−9610.

